

UNCLASSIFIED



3 4456 0133949 9

CENTRAL COLLECTION

ORNL-2192  
Chemistry

137

OPTICAL PROPERTIES AND X-RAY DIFFRACTION  
DATA FOR SOME INORGANIC FLUORIDE  
AND CHLORIDE COMPOUNDS

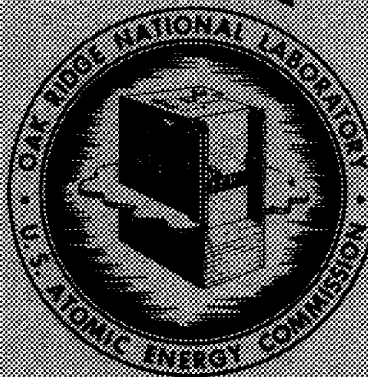
H. Insley  
T. N. McVay  
R. E. Thoma  
G. D. White

CENTRAL RESEARCH LIBRARY  
DOCUMENT COLLECTION

**LIBRARY LOAN COPY**

DO NOT TRANSFER TO ANOTHER PERSON

If you wish someone else to see this document  
send in name with document and the library will  
arrange a loan.



**OAK RIDGE NATIONAL LABORATORY**

OPERATED BY

**UNION CARBIDE NUCLEAR COMPANY**

A Division of Union Carbide and Carbon Corporation



POST OFFICE BOX P • OAK RIDGE, TENNESSEE

UNCLASSIFIED

Printed in USA. Price 45 cents. Available from the

Office of Technical Services  
U. S. Department of Commerce  
Washington 25, D. C.

#### LEGAL NOTICE

This report was prepared as an account of Government sponsored work. Neither the United States, nor the Commission, nor any person acting on behalf of the Commission:

- A. Makes any warranty or representation, express or implied, with respect to the accuracy, completeness, or usefulness of the information contained in this report, or that the use of any information, apparatus, method, or process disclosed in this report may not infringe privately owned rights; or
- B. Assumes any liabilities with respect to the use of, or for damages resulting from the use of any information, apparatus, method, or process disclosed in this report.

As used in the above, "person acting on behalf of the Commission" includes any employee or contractor of the Commission to the extent that such employee or contractor prepares, handles or distributes, or provides access to, any information pursuant to his employment or contract with the Commission.

UNCLASSIFIED

ORNL-2192

Contract No. W-7405-eng-26

CHEMISTRY DIVISION

and

METALLURGY DIVISION

OPTICAL PROPERTIES AND X-RAY DIFFRACTION DATA  
FOR SOME INORGANIC FLUORIDE AND CHLORIDE COMPOUNDS

H. Insley, Consultant  
T. N. McVay, Consultant  
R. E. Thoma, Chemistry Division  
G. D. White, Metallurgy Division

Date Issued

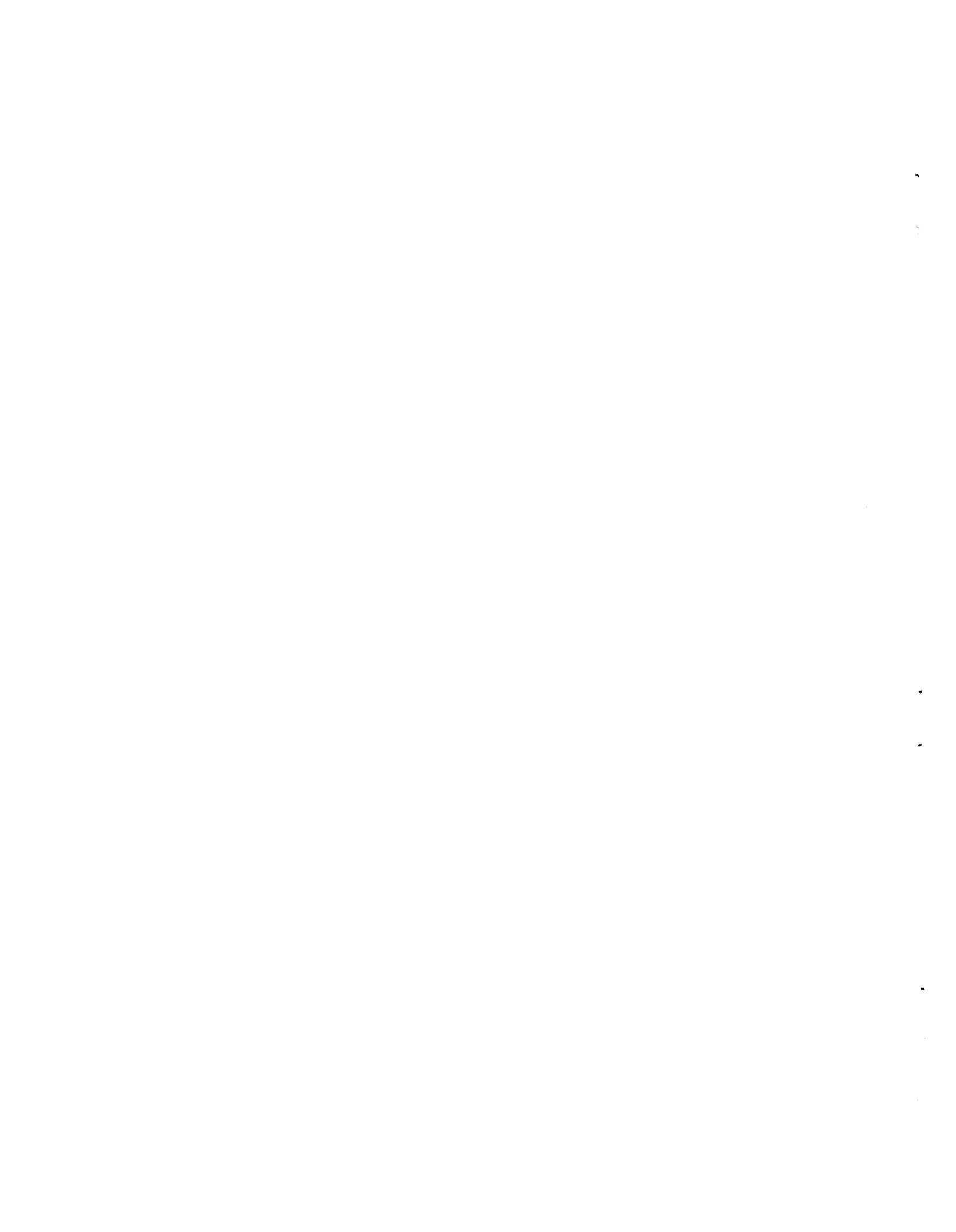
OCT 23 1958

OAK RIDGE NATIONAL LABORATORY  
Operated by  
UNION CARBIDE NUCLEAR COMPANY  
A Division of Union Carbide and Carbon Corporation  
Oak Ridge, Tennessee

UNCLASSIFIED



3 4456 0133949 9





INTERNAL DISTRIBUTION

1. R. G. Affel
2. C. J. Barton
3. M. Bender
4. D. S. Billington
5. F. F. Blankenship
6. E. P. Blizard
7. C. J. Borkowski
8. W. F. Boudreau
9. G. E. Boyd
10. M. A. Bredig
11. W. E. Browning
12. F. R. Bruce
13. A. D. Callihan
14. D. W. Cardwell
15. C. E. Center (K-25)
16. R. A. Charpie
17. C. E. Clifford
18. J. H. Coobs
19. W. B. Cottrell
20. D. D. Cowen
21. S. J. Cromer
22. R. S. Crouse
23. F. L. Culler
24. J. H. DeVan
25. L. M. Doney
26. D. A. Douglas
27. E. R. Dytko
28. W. K. Eister
29. L. B. Emlet (K-25)
30. D. E. Ferguson
31. A. P. Fraas
32. J. H. Frye, Jr.
33. W. T. Furgerson
34. H. C. Gray
35. W. R. Grimes
36. E. E. Hoffman
37. H. W. Hoffman
38. A. Hollaender
39. A. S. Householder
40. J. T. Howe
41. W. H. Jordan
42. G. W. Keilholtz
43. C. P. Keim
44. M. T. Kelley
45. F. Kertesz
46. E. M. King
47. J. A. Lane
48. R. B. Lindauer
49. R. S. Livingston
50. R. N. Lyon
51. F. C. Maienschein
52. W. D. Manly
53. E. R. Mann
54. L. A. Mann
55. W. B. McDonald
56. F. R. McQuilkin
57. R. V. Meghreblian
58. R. P. Milford
59. A. J. Miller
60. R. E. Moore
61. J. G. Morgan
62. K. Z. Morgan
63. E. J. Murphy
64. J. P. Murray (Y-12)
65. M. L. Nelson
66. G. J. Nettle
67. R. B. Oliver
68. L. G. Overholser
69. P. Patriarca
70. R. W. Peelle
71. A. M. Perry
72. J. C. Pigg
73. H. F. Poppendiek
74. A. E. Richt
75. M. T. Robinson
76. H. W. Savage
77. A. W. Savolainen
78. R. D. Schultheiss
79. E. D. Shipley
80. A. Simon
81. O. Sisman
82. J. Sites
83. M. J. Skinner
84. G. P. Smith
85. A. H. Snell
86. C. D. Susano
87. J. A. Swartout
88. E. H. Taylor
- 89-91. R. E. Thoma
92. D. B. Trauger
93. E. R. Van Artsdalen
94. G. M. Watson
95. A. M. Weinberg
96. J. C. White

- |                                   |   |
|-----------------------------------|---|
| 97. G. D. Whitman                 | 109-111. G. D. White  |
| 98. E. P. Wigner (consultant)     | 112. H. L. Yaskel, Jr.  |
| 99. J. C. Wilson                  | 113-114. ORNL - Y-12 Technical Library,<br>Document Reference Section |
| 100. C. E. Winters                | 115-134. Laboratory Records Department                                |
| 101. P. A. Agron                  | 135. Laboratory Records, ORNL R.C.                                    |
| 102-104. T. N. McVay (consultant) | 136. Metallurgy Library   |
| 105-107. H. Insley (consultant)   | 137-138. Central Research Library                                     |
| 108. R. M. Steele                 |   |

EXTERNAL DISTRIBUTION

- 139. R. F. Bacher, California Institute of Technology
- 140. Division of Research and Development, AEC, ORO
- 141-725. Given distribution as shown in TID-4500 under Chemistry category  
(200 copies - OTS)

DISTRIBUTION PAGE TO BE REMOVED IF REPORT IS GIVEN PUBLIC DISTRIBUTION

OPTICAL PROPERTIES AND X-RAY DIFFRACTION DATA  
FOR SOME INORGANIC FLUORIDE AND CHLORIDE COMPOUNDS

H. Insley, Consultant  
T. N. McVay, Consultant  
R. E. Thoma, Chemistry Division  
G. D. White, Metallurgy Division

ABSTRACT

Optical properties and X-ray diffraction data are listed for various inorganic fluoride and chloride compounds. This publication extends and replaces ORNL-1712, Properties of Some Inorganic Fluoride and Chloride Compounds, by T. N. McVay and G. D. White.

INTRODUCTION

Optical and X-ray diffraction data have been collected for many compounds whose existence was not known prior to their discovery incidental to the phase equilibrium studies made in the High Temperature Section of the Chemistry Division. A few compounds are listed and mentioned in the footnotes whose initial discovery was not made at ORNL; however, the original optical measurements on these compounds were made by H. Insley, T. N. McVay, and G. D. White of the Ceramics Laboratory, Metallurgy Division.

The standard X-ray diffraction patterns included herein were derived, in general, from the same samples on which the optical data were taken. Standard patterns were made from powder samples with a Norelco-Phillips high angle diffractometer, using Cu K $\alpha$  filtered radiation from a General Electric CA-7 X-ray tube. The diffractometer was equipped with a Geiger Muller tube counting arrangement. The X-ray data have not been corrected for absorption. Values for interplanar distances (d, measured in Angstrom units) and relative intensities of diffracting maxima conform to the conventions used in the ASTM X-ray diffraction data cards.

The refractive indices of the compounds included are believed to be precise to  $\pm 0.003$ ; the optic angles of biaxial crystals were estimated.

This publication is divided into Part I (optical properties) and Part II (X-ray diffraction data). The three strongest X-ray lines accompany the optical data of each compound for which ASTM X-ray diffraction data are not available, if those X-ray data were derived at ORNL. X-ray diffraction data in Part II are separated according to the purity of the samples used for standards. Patterns for the compounds in the first section were derived from single-phase samples of known chemical analysis which had met optical standards. In the second section are listed data on compounds whose purity has not been absolutely established.

#### ACKNOWLEDGEMENTS

The authors wish to acknowledge the assistance of L. M. Bratcher, V. S. Coleman, H. A. Friedman, R. J. Sheil, B. J. Sturm, J. Truitt, and W. C. Whitley, who prepared the samples under the direction of C. J. Barton, F. F. Blankenship, R. E. Moore and W. R. Grimes.



PART I

OPTICAL PROPERTIES



Ammonium beryllium fluoride,  $2 \text{NH}_4 \cdot \text{BeF}_2$

Uniaxial +  
 $n = 1.319$   
Low birefringence  
Colorless  
Questionable. National Research Council Bulletin 118  
describes it as rhombic.

Beryllium fluoride,  $\text{BeF}_2$

Uniaxial +  
 $n = 1.325$   
Low birefringence estimated .006  
Quartz form  
Colorless  
Sample prepared at Mound Laboratory.

Beryllium lead fluoride,  $\text{BeF}_2 \cdot \text{PbF}_2$

Biaxial -  $2V = 70^\circ$   
 $\alpha = 1.602$   $\gamma = 1.627$   
Colorless

Cesium beryllium fluoride,  $2 \text{CsF} \cdot \text{BeF}_2$

$n = 1.452$   
Very low birefringence  
Colorless.

Cesium beryllium fluoride,  $\text{CsF} \cdot \text{BeF}_2$

$n = 1.382$   
Low birefringence  
Length slow. Parallel extinction.  
Colorless.

Cesium lanthanum fluoride,  $3 \text{CsF} \cdot \text{LaF}_3$

Cubic  
 $n = 1.462$   
Colorless  
X-ray lines: 3.49, 2.478, 2.021.

Cesium uranium fluoride,  $\text{CsF} \cdot \text{UF}_4$

Biaxial +  $2V = 45^\circ$   
 $\alpha = 1.553$   $\gamma = 1.560$   
Polysynthetic twinning  $X_{\Delta c} = 10^\circ$   $Z = \text{sky blue}$   
 $X = \text{greenish blue}$   
X-ray lines: 7.31, 4.00, 3.62

Cesium uranium fluoride,  $2 \text{CsF} \cdot \text{UF}_4$

Biaxial +  $2V = 45^\circ$   
 $\alpha = 1.516$   $\gamma = 1.524$   
X = light greenish blue Z = light blue  
X-ray lines: 6.19, 3.55, 3.44.

Cesium zinc fluoride,  $2 \text{CsF} \cdot \text{ZnF}_2$

Biaxial +  $2V = \text{small}$   
 $\alpha = 1.446$   $\gamma = 1.458$   
Colorless

Cesium zirconium fluoride,  $\text{CsF} \cdot \text{ZrF}_4$

Biaxial -  $2V = 20^\circ - 45^\circ$  (varies)  
 $\alpha = 1.464$   $\gamma = 1.476$   
Colorless  
X-ray lines: 3.73, 3.62, 3.26

Cesium zirconium fluoride,  $2 \text{CsF} \cdot \text{ZrF}_4$

Uniaxial -  $\epsilon = 1.460$   
 $\omega = 1.482$   
Colorless  
X-ray lines: 3.73, 3.20, 2.430.

Chromium fluoride,  $\text{CrF}_2$

Biaxial +  $2V = 10^\circ$   
Monoclinic  $X_{Ac} = 38^\circ$   
 $\alpha = 1.511$   $\gamma = 1.525$   
Gray green  
X-ray lines: 3.53, 2.97, 2.805

Iron fluoride,  $\text{FeF}_2$

Uniaxial +  $\epsilon = 1.540$   
 $\omega = 1.524$   
Brown

Iron zirconium fluoride,  $\text{FeF}_2 \cdot \text{ZrF}_4$

Cubic  
 $n = 1.432$   
X-ray lines: 4.04, 2.016, 1.805



Lanthanum fluoride,  $\text{LaF}_3$

Uniaxial -  
Hexagonal  
 $\omega = 1.605$   $\epsilon = 1.594$   
Length fast  
Prismatic  
Colorless.

Lanthanum zirconium uranium fluoride,  $\text{LaF}_3 \cdot 6\text{ZrF}_4 \cdot \text{UF}_4$

Biaxial +  $2V = 70^\circ$   
 $\alpha = 1.528$   $\gamma = 1.545$   
Light green.

Lead uranium fluoride,  $\text{PbF}_2 \cdot \text{UF}_4$

Uniaxial -  
 $\omega = 1.750$   $\epsilon = 1.730$   
Green.

Lead uranium fluoride,  $6 \text{PbF}_2 \cdot \text{UF}_4$

Cubic  
 $n = 1.77$   
Light blue.

Lithium beryllium fluoride,  $2 \text{LiF} \cdot \text{BeF}_2$

Uniaxial +  
 $\omega = 1.312$   $\epsilon = 1.319$   
Colorless.

Lithium cesium fluoride,  $\text{LiF} \cdot \text{CsF}$

Biaxial +  $2V = \text{small}$   
 $n = 1.458$   
Estimated birefringence 0.006  
Colorless

Lithium chromium fluoride,  $3 \text{LiF} \cdot \text{CrF}_3$

Biaxial -  $2V = 40^\circ$   
 $\alpha = 1.444$   $\gamma = 1.464$   
Green  
X-ray lines: 4.29, 4.16, 2.176.

Lithium rubidium fluoride,  $\text{LiF}\cdot\text{RbF}$

Biaxial +  
Orthorhombic  
 $n = 1.396$   
Low birefringence  
Colorless.

Lithium sodium beryllium fluoride,  $\text{LiF}\cdot 2\text{NaF}\cdot 2\text{BeF}_2$

Uniaxial -  
 $n = 1.311$   
Low birefringence  
Colorless.

Lithium uranium fluoride,  $4\text{LiF}\cdot\text{UF}_4$

Biaxial +  $2V = 45^\circ$   
 $\alpha = 1.460$   $\gamma = 1.472$   
X = light green Z = dark green  
X-ray lines: 5.13, 4.93, 4.44.

Lithium uranium fluoride,  $7\text{LiF}\cdot 6\text{UF}_4$

Uniaxial -  
 $\omega = 1.554$   $\epsilon = 1.551$   
Deep green  
X-ray lines: 5.24, 3.33, 2.99.

Lithium uranium fluoride,  $\text{LiF}\cdot 4\text{UF}_4$

Biaxial -  $2V = 10^\circ$   
 $\alpha = 1.584$   $\gamma = 1.600$   
Yellowish green  
X-ray lines: 4.25, 3.78, 3.52.

Lithium zirconium fluoride,  $\text{LiF}\cdot\text{ZrF}_4$

Biaxial +  $2V = 30^\circ$   
 $\alpha = 1.468$   $\gamma = 1.476$   
Colorless  
Quench determinations show this phase to have  
compositions  $3\text{LiF}\cdot 4\text{ZrF}_4$  with biaxial +.  
X-ray lines: 3.90, 3.33, 3.16.

Lithium zirconium fluoride,  $3\text{LiF}\cdot 4\text{ZrF}_4$

Biaxial +  $2V = 30^\circ$   
 $\alpha = 1.463$   $\gamma = 1.473$   
Colorless  
X-ray lines: 3.90, 3.33, 3.16.

Lithium zirconium fluoride, 2 LiF·ZrF<sub>4</sub>

Uniaxial +  
 $\omega = 1.468$   $\epsilon = 1.478$   
Colorless  
X-ray lines: 4.29, 3.15, 2.19

Lithium zirconium fluoride, 3 LiF·ZrF<sub>4</sub>

Biaxial -  $2V = 30^\circ$   
 $\alpha = 1.445$   $\gamma = 1.465$   
Colorless  
X-ray lines: 5.49, 4.88, 2.07

Manganese fluoride, MnF<sub>2</sub>

Uniaxial +  
 $\omega = 1.476$   $\epsilon = 1.504$   
X = colorless  $Z = \text{gray}$

Nickel fluoride, NiF<sub>2</sub>

Uniaxial +  
 $\omega = 1.526$   $\epsilon = 1.560$   
Light greenish yellow.

Nickel zirconium fluoride, NiF<sub>2</sub>·ZrF<sub>4</sub>

$n = 1.442$   
Very low birefringence  
X-ray lines: 3.91, 1.964, 1.767.

Potassium aluminum fluoride, 3 KF·AlF<sub>3</sub>

Cubic  
 $n = 1.376$   
Colorless.

Potassium beryllium fluoride, 3 KF·BeF<sub>2</sub>

Uniaxial +  
 $\omega = 1.357$   $\epsilon = 1.366$   
Colorless  
X-ray lines: 2.98, 2.39, 2.27.

Potassium beryllium fluoride, 2KF·BeF<sub>2</sub>

Biaxial +  $2V = 30^\circ$   
 $\alpha = 1.357$   $\gamma = 1.366$   
Probably monoclinic  
 $X_{\Delta c} = 40^\circ$   
Colorless.

Potassium beryllium fluoride,  $\text{KF}\cdot\text{BeF}_2$

Biaxial -  $2V = 45^\circ$   
 $\alpha = 1.319$   $\gamma = 1.323$   
Colorless  
X-ray lines: 5.99, 3.33, 3.01.

Potassium beryllium fluoride,  $\text{KF}\cdot 2\text{BeF}_2$

Uniaxial -  
 $\omega = 1.319$   $\epsilon = 1.312$   
Colorless  
X-ray lines: 3.31, 3.00, 2.29.

Potassium chromium fluoride,  $3 \text{KF}\cdot\text{CrF}_3$

Cubic  
 $n = 1.422$   
Green  
X-ray lines: 4.95, 3.03, 2.35.

Potassium fluoride acid,  $\text{KF}\cdot\text{HF}$

Uniaxial -. Some crystals have a small optic angle.  
 $\omega = 1.354$   $\epsilon = 1.331$

Potassium iron chloride,  $\text{KCl}\cdot\text{FeCl}_2$

Biaxial +  $2V = 20^\circ$   
 $\alpha = 1.700 \pm 0.005$   $\gamma = 1.740 \pm 0.005$   
Colorless.

Potassium iron chloride,  $2 \text{KCl}\cdot\text{FeCl}_2$

Uniaxial +  
 $\omega = 1.600$   $\epsilon = 1.636$   
Colorless.

Potassium lanthanum fluoride,  $\text{KF}\cdot\text{LaF}_3$

Uniaxial +  
 $\omega = 1.493$   $\epsilon = 1.510$   
Colorless  
X-ray lines: 7.77, 2.747, 1.857.

Potassium nickel fluoride,  $2 \text{KF}\cdot\text{NiF}_2$

Uniaxial -  
Tetragonal  
 $\omega = 1.434$   $\epsilon = 1.426$   
Yellow  
X-ray lines: 6.51, 2.94, 2.173.



Potassium sodium beryllium fluoride,  $\text{KF}\cdot\text{NaF}\cdot\text{BeF}_2$

Biaxial + 2V = small  
n = 1.343  
Low birefringence  
Colorless  
X-ray lines: 2.786, 2.367, 1.995.

Potassium sodium iron fluoride,  $2 \text{KF}\cdot\text{NaF}\cdot\text{FeF}_3$

Cubic  
n = 1.414  
Colorless

Potassium sodium zirconium fluoride,  $3 \text{KF}\cdot 3\text{NaF}\cdot 2\text{ZrF}_4$

Biaxial + 2V = 30°  
 $\alpha = 1.410$   $\gamma = 1.421$   
Colorless  
X-ray lines: 4.90, 4.09, 2.97.

Potassium sodium zirconium fluoride,  $\text{KF}\cdot\text{NaF}\cdot\text{ZrF}_4$

Biaxial - 2V = 60°  
 $\alpha = 1.378$   $\gamma = 1.385$   
Colorless  
Often fibrous or prismatic  
X-ray lines: 4.85, 4.09, 3.34

Potassium sodium zirconium fluoride,  $3\text{KF}\cdot 2\text{NaF}\cdot 5\text{ZrF}_4$

Biaxial - 2V = about 80°  
 $\alpha = 1.478$   $\gamma = 1.488$   
Colorless  
Probably monoclinic  
Marked cleavage; Z $\Delta$  elong. about 7°  
X-ray lines: 8.51, 7.70, 3.86.

Potassium tellurium fluoride,  $\text{KTeF}_5\cdot\text{H}_2\text{O}$

Biaxial + 2V = 30°  
 $\alpha = 1.436$   $\gamma = 1.460$   
Colorless  
X-ray lines: 5.64, 4.27, 3.575.

Potassium thorium fluoride,  $3 \text{KF}\cdot\text{ThF}_4$

Cubic  
n = 1.424  
Colorless

Potassium uranium chloride,  $\text{KCl} \cdot \text{UCl}_4$

Biaxial +  $2V = \text{small}$   
 $\alpha = 1.692$   $\beta = 1.705$   $\gamma = 1.759$   
X = gray Z = blue green

Potassium uranium fluoride,  $7 \text{KF} \cdot 6 \text{UF}_4$

Uniaxial -  
 $\omega = 1.510$   $\epsilon = 1.504$   
Green  
X-ray lines: 4.82, 3.44, 1.793

Potassium uranium fluoride,  $\text{KF} \cdot \text{UF}_3$

Isomorphous with  $\text{NaF} \cdot \text{UF}_3$   
Similar optical properties.

Potassium uranium fluoride,  $\text{KF} \cdot 2\text{UF}_4$

Biaxial -  $2V = 15^\circ$   
 $\alpha = 1.520$   $\gamma = 1.584$   
Green.

Potassium uranium fluoride,  $2 \text{KF} \cdot \text{UF}_4$

Uniaxial +  
 $\omega = 1.484$   $\epsilon = 1.512$   
Light olive drab.

Potassium zinc fluoride,  $\text{KF} \cdot \text{ZnF}_2$

Cubic  
 $n = 1.462$   
Colorless.

Potassium zinc fluoride,  $2 \text{KF} \cdot \text{ZnF}_2$

Uniaxial -  
 $\omega = 1.416$   $\epsilon = 1.410$

Potassium zirconium fluoride,  $2 \text{KF} \cdot \text{ZrF}_4$

"Low temperature" form  
Biaxial +  $2V = 10^\circ$   
 $\alpha = 1.412$   $\gamma = 1.424$   
Polysynthetic twinning frequent  
Colorless.

Potassium zirconium fluoride,  $\alpha\text{KF}\cdot\text{ZrF}_4$

Biaxial -  $2V = 80^\circ$   
 $\alpha = 1.412$   $\gamma = 1.428$   
Colorless  
X-ray lines: 5.83, 4.40, 3.86.

Potassium zirconium fluoride,  $\beta\text{KF}\cdot\text{ZrF}_4$

Biaxial +  $2V = 75^\circ$   
 $\alpha = 1.488$   $\gamma = 1.504$   
Colorless  
X-ray lines: 6.32, 3.29, 3.15.

Rubidium beryllium fluoride,  $\text{RbF}\cdot 2\text{BeF}_2$

Biaxial -  $2V = 70^\circ$   
 $n = 1.333$   
Birefringence about 0.008  
Colorless

Rubidium cerium fluoride,  $\text{RbF}\cdot\text{CeF}_3$

Biaxial +  $2V = 75^\circ$   
 $\alpha = 1.500$   $\gamma = 1.520$

Rubidium lanthanum fluoride,  $\text{RbF}\cdot\text{LaF}_3$

Biaxial +  $2V = 70^\circ$   
 $\alpha = 1.498$   $\gamma = 1.519$   
Colorless  
X-ray lines: 3.43, 2.25, 1.90.

Rubidium lanthanum fluoride,  $3\text{RbF}\cdot\text{LaF}_3$

Cubic  
 $n = 1.424$   
Colorless  
X-ray lines: 3.30, 2.821, 3.11.

Rubidium sodium beryllium fluoride,  $\text{RbF}\cdot 2\text{NaF}\cdot\text{BeF}_2$

Cubic  
 $n = 1.374$   
Colorless  
X-ray lines: 2.99, 2.875, 2.494.

Rubidium sodium uranium fluoride,  $\text{RbF}\cdot\text{NaF}\cdot\text{UF}_4$

Uniaxial +  
 $\omega = 1.484$   $\epsilon = 1.486$   
Anomalous purple interference color characteristic.  
Pale green  
X-ray lines: 8.12, 3.26, 2.034.

Rubidium uranium fluoride, 3 RbF·UF<sub>4</sub>

Cubic  
n = 1.438  
Green  
X-ray lines: 5.56, 3.40, 1.959.

Rubidium uranium fluoride, RbF·UF<sub>4</sub>

Biaxial - 2V = 75°  
 $\alpha = 1.514$   $\gamma = 1.528$   
Polysynthetic twinning  
Y<sub>Ac</sub> = 20°  
Lath-shaped crystals  
X = green Z = blue  
Marked dispersion of optic axes  
X-ray lines: 6.86, 3.46, 3.43.

Rubidium uranium fluoride, 2 RbF·UF<sub>4</sub>

Biaxial + 2V = 70°  
 $\alpha = 1.473$   $\gamma = 1.487$   
X = light green Z = light violet  
Polysynthetic twinning common  
X-ray lines: 6.03; 3.49, 2.008.

Rubidium uranium fluoride, 7 RbF·6 UF<sub>4</sub>

Uniaxial -  
 $\omega = 1.518$   $\epsilon = 1.512$   
X-ray lines: 3.51, 2.140, 1.835.

Rubidium uranium fluoride, 2 RbF·3 UF<sub>4</sub>

Biaxial - 2V = 60°  
 $\alpha = 1.542$   $\gamma = 1.550$   
Inclined extinction  
Pleochroic, bluish green and pale violet-green  
X-ray lines: 5.80, 3.50, 2.038.

Rubidium uranium fluoride, RbF·3UF<sub>4</sub>

Biaxial + 2V = 70°  
 $\alpha = 1.588$   $\gamma = 1.598$   
Yellow green with slight dichroism  
Strong dispersion of optic axes  
Anomalous dispersion of birefringence in purplish red  
X-ray lines: 3.52, 3.36, 2.002.



Rubidium uranium fluoride,  $\text{RbF} \cdot 6\text{UF}_4$

Uniaxial -  
Probably tetragonal  
 $\omega = 1.596$   $\epsilon = 1.586$   
Deep green  
X-ray lines: 4.13, 3.47, 2.049

III

Rubidium uranium fluoride,  $3\text{RbF} \cdot \text{UF}_3$

Cubic  
 $n = 1.440$  approximate  
Pinkish brown.

Rubidium zirconium fluoride,  $3\text{RbF} \cdot \text{ZrF}_4$

Cubic  
 $n = 1.420$   
Colorless  
X-ray lines: 3.290, 2.236, 1.990

Rubidium zirconium fluoride,  $2 \text{RbF} \cdot \text{ZrF}_4$

Uniaxial - (low temperature form)  
 $\omega = 1.440$   $\epsilon = 1.426$   
Colorless  
X-ray lines: 3.590, 3.080, 2.338.

Rubidium zirconium fluoride,  $5 \text{RbF} \cdot 4\text{ZrF}_4$

Biaxial -  $2V$  about  $85^\circ$   
 $\alpha = 1.442$   $\gamma = 1.452$   
Colorless  
X-ray lines: 3.44, 3.36, 3.29.

Rubidium zirconium fluoride,  $\text{RbF} \cdot \text{ZrF}_4$

Biaxial -  $2V =$  about  $75^\circ$   
 $\alpha = 1.490$   $\gamma = 1.502$   
Elong. fast extinction about  $2^\circ$   
X-ray lines: 3.42, 3.36, 3.325.

Sodium beryllium fluoride,  $\text{NaF} \cdot \text{BeF}_2$

Biaxial  $2V =$  large  
 $n = 1.312$   
Low birefringence  
Length slow  
colorless

Sodium beryllium fluoride,  $2 \text{ NaF} \cdot \text{BeF}_2$

Low temperature  
Biaxial  
 $n = 1.303$   
Low birefringence  
Yllc  
Colorless  
Sometimes twinned.

Sodium beryllium fluoride,  $2 \text{ NaF} \cdot \text{BeF}_2$

High temperature form.  
Cubic  
 $n = 1.333$   
Colorless.

Sodium cerium fluoride,  $\text{NaF} \cdot \text{CeF}_3$

Uniaxial +  
 $\omega = 1.493$   $\epsilon = 1.514$   
X-ray lines: 3.09, 2.57, 1.777.

Sodium chromium fluoride,  $3 \text{ NaF} \cdot \text{CrF}_3$

Cubic  
 $n = 1.411$   
Green

Sodium fluoride acid,  $\text{NaF} \cdot \text{HF}$

Uniaxial +  
 $\omega = 1.261$   $\epsilon = 1.328$   
Colorless

Sodium lanthanum fluoride,  $\text{NaF} \cdot \text{LaF}_3$

Uniaxial +  
 $\omega = 1.486$   $\epsilon = 1.500$   
X-ray lines: 3.11, 3.09, 2.194.

Sodium thorium fluoride,  $2 \text{ NaF} \cdot \text{ThF}_4$

Uniaxial +  
 $\omega = 1.464$   $\epsilon = 1.496$   
Colorless  
X-ray lines: 5.17, 2.99, 1.742

Sodium uranium chloride,  $2 \text{NaCl} \cdot \text{UCl}_4$

Uniaxial -  
 $\omega = 1.664$   $\epsilon = 1.652$   
Pale green

Sodium uranium fluoride,  $\text{NaF} \cdot \text{UF}_3$

Uniaxial +  
 $\omega = 1.552$   $\epsilon = 1.564$   
Dark blue  
X-ray lines: 3.09, 2.179, 1.779.

Sodium uranium fluoride,  $2 \text{NaF} \cdot \text{UF}_4$

Uniaxial -  
 $\omega = 1.495$   $\epsilon = 1.490$   
Green  
X-ray lines: 7.25, 4.28, 2.99.

Sodium uranium fluoride,  $\beta 2$ ,  $2\text{NaF} \cdot \text{UF}_4$

Uniaxial +  
 $\omega = 1.484$   $\epsilon = 1.522$   
X = greenish-tan  $Z = \text{gray-tan}$

Sodium uranium fluoride,  $3\text{NaF} \cdot \text{UF}_4$

Uniaxial -  
 $\omega = 1.417$   $\epsilon = 1.411$   
Greenish blue  
X-ray lines: 5.15, 2.97, 2.102.

Sodium uranium fluoride,  $5\text{NaF} \cdot 3\text{UF}_4$

Cubic  
 $n = 1.475$   
Green  
X-ray lines: 3.19, 1.964, 1.684.

Sodium uranium fluoride,  $\text{NaF} \cdot 2\text{UF}_4$

Biaxial -  $2V = 60^\circ$   
Orthorhombic  
 $\alpha = 1.516$   $\gamma = 1.584$   
Yellowish green  
X-ray lines: 5.61, 3.26, 3.08.

Sodium uranium fluoride, 7 NaF·6 UF<sub>4</sub>

Uniaxial -  
 $\omega = 1.520$   $\epsilon = 1.512$   
Green  
X-ray lines: 4.32, 3.33, 3.25.

Sodium zinc fluoride, 2 NaF·ZnF<sub>2</sub>

Uniaxial -  
Tetragonal  
 $\omega = 1.418$   $\epsilon = 1.410$   
Colorless

Sodium zirconium fluoride, 3 NaF·4 ZrF<sub>4</sub>

Biaxial +  $2V = 30^\circ$   
 $\alpha = 1.420$   $\gamma = 1.432$   
Colorless  
X-ray lines: 4.15, 3.36, 2.074.

Sodium zirconium fluoride, NaF·ZrF<sub>4</sub>

Metastable phase  
Uniaxial +  
 $\omega = 1.417$   $\epsilon = 1.446$   
Colorless  
X-ray lines: 3.37, 3.86, 2.09.

Sodium zirconium fluoride, 7 NaF·6 ZrF<sub>4</sub>

Uniaxial -  
 $\omega = 1.508$   $\epsilon = 1.500$   
Indices depend on composition  
Colorless  
Some solid solution in this area  
X-ray lines: 3.13, 1.92, 1.91.

Sodium zirconium fluoride, 2 NaF·ZrF<sub>4</sub>

" $\beta_1$ " form  
Uniaxial +  
 $\omega = 1.406$   $\epsilon = 1.408$   
Colorless  
Partially miscible with 3 NaF·ZrF<sub>4</sub>  
X-ray lines: 5.15, 3.09, 1.890.

Sodium zirconium fluoride, 2 NaF·ZrF<sub>4</sub>

Biaxial -  $2V = 75^\circ$   
 $\alpha = 1.412$   $\gamma = 1.419$   
Colorless  
"β<sub>2</sub>" form  
X-ray lines: 5.47, 3.11, 1.912.

Sodium zirconium fluoride, 2 NaF·ZrF<sub>4</sub>

"β<sub>3</sub>" form, stable 500-530° approx.  
Uniaxial +  
 $\omega = 1.376$   $\epsilon = 1.386$   
Colorless  
X-ray lines: 4.55, 2.893, 1.894.

Sodium zirconium fluoride, 2 NaF·ZrF<sub>4</sub>

"β<sub>4</sub>" form  
Biaxial +  $2V = 75^\circ$   
 $\alpha = 1.408$   $\gamma = 1.412$   
Colorless  
X-ray lines: 5.12, 3.83, 3.25.

Sodium zirconium fluoride, 2 NaF·ZrF<sub>4</sub>

"γ" form  
Biaxial -  $2V > 70^\circ$   
 $\alpha = 1.420$   $\gamma = 1.429^\circ$   
Colorless  
Polysynthetic twinning common  
X-ray lines: 4.98, 4.19, 3.07.

Sodium zirconium fluoride, 3 NaF·ZrF<sub>4</sub>

Uniaxial -  
 $\omega = 1.386$   $\epsilon = 1.381$   
Colorless  
X-ray lines: 4.75, 3.06, 1.87.

Thorium fluoride, ThF<sub>4</sub>

Biaxial -  $2V = 60^\circ$   
 $\alpha = 1.500$   $\gamma = 1.534$   
Colorless

Uranium chloride,  $\text{UCl}_3$

Uniaxial probably -  
High  $n$  2.04  
Dark brownish red

Low  $n$  1.94

Uranium chloride,  $\text{UCl}_4$

Uniaxial -  
 $\omega = 2.03$   
 $X =$  light brownish green

$\epsilon = 1.95$   
 $Z =$  greenish brown

Yttrium fluoride,  $\text{YF}_3$

Biaxial  
 $\alpha = 1.536$   
Colorless

$2V \sim 90^\circ$   
 $\gamma = 1.568$

Zinc fluoride,  $\text{ZnF}_2$

Uniaxial +  
 $\omega = 1.501$   
Colorless

$\epsilon = 1.526$

Zinc zirconium fluoride,  $\text{ZnF}_2 \cdot \text{ZrF}_4$

Cubic  
 $n = 1.434$   
Colorless.

Zirconium chloride,  $\text{ZrCl}_4$

Probably monoclinic  
Biaxial  
 $\alpha = 1.76$   
 $Z\Delta c = 22^\circ$   
Colorless

$2V =$  Large  
 $\gamma = 1.83$

Zirconium uranium fluoride,  $3 \text{ZrF}_4 \cdot \text{UF}_3$

Biaxial  
 $n = 1.560$  approximate average  
Red  
X-ray lines: 4.11, 3.72, 2.06

$2V =$  Large

PART II

X-RAY DIFFRACTION DATA





A. X-ray diffraction patterns for the compounds listed below are included in this section:

BeF <sub>2</sub>	3KF·2NaF·5ZrF <sub>4</sub>	RbF·NaF·2ZrF <sub>4</sub>
2CsF·BeF <sub>2</sub>	KF·NaF·ZrF <sub>4</sub>	RbF·NaF·ZrF <sub>4</sub>
CsF·BeF <sub>2</sub>	2KF·NiF <sub>2</sub>	NaF·CeF <sub>3</sub>
3CsF·LaF <sub>3</sub>	KF·TeF <sub>4</sub>	NaF·CrF <sub>2</sub>
2CsF·UF <sub>4</sub>	7KF·6UF <sub>4</sub>	3NaF·HfF <sub>4</sub>
CsF·UF <sub>4</sub>	3KF·2ZrF <sub>4</sub>	NaF·HfF <sub>4</sub>
2CsF·ZrF <sub>4</sub>	α-KF·ZrF <sub>4</sub>	NaF·FeF <sub>2</sub>
CsF·ZrF <sub>4</sub>	β-KF·ZrF <sub>4</sub>	NaF·LaF <sub>3</sub>
CrF <sub>2</sub>	3RbF·BeF <sub>2</sub>	NaF·2LiF·CrF <sub>3</sub>
CrF <sub>3</sub>	2RbF·BeF <sub>2</sub>	NaF·NiF <sub>2</sub>
CrF <sub>2</sub> ·ZrF <sub>4</sub>	RbF·BeF <sub>2</sub>	2NaF·ThF <sub>4</sub>
FeF <sub>2</sub> ·ZrF <sub>4</sub>	RbF·2BeF <sub>2</sub>	NaF·UF <sub>3</sub>
3LiF·CrF <sub>3</sub>	3RbF·CrF <sub>3</sub>	α-3NaF·UF <sub>4</sub>
3LiF·NiF <sub>2</sub>	RbF·LaF <sub>3</sub>	β-3NaF·UF <sub>4</sub>
4LiF·UF <sub>4</sub>	RbF·2NaF·BeF <sub>2</sub>	β-2 2NaF·UF <sub>4</sub>
3LiF·UF <sub>4</sub>	RbF·NaF·UF <sub>4</sub>	β-3 2NaF·UF <sub>4</sub>
7LiF·6UF <sub>4</sub>	3RbF·3NaF·2ZrF <sub>4</sub>	γ 2NaF·UF <sub>4</sub>
LiF·4UF <sub>4</sub>	2RbF·UF <sub>4</sub>	5NaF·3UF <sub>4</sub>
3LiF·ZrF <sub>4</sub>	2RbF·UF <sub>4</sub>	7NaF·6UF <sub>4</sub>
2LiF·ZrF <sub>4</sub>	7RbF·6UF <sub>4</sub>	NaF·2UF <sub>4</sub>
3LiF·4ZrF <sub>4</sub>	RbF·UF <sub>4</sub>	3NaF·ZrF <sub>4</sub>
NiF <sub>2</sub> ·ZrF <sub>4</sub>	2RbF·3UF <sub>4</sub>	β-1 2NaF·ZrF <sub>4</sub>
3KF·BeF <sub>2</sub>	RbF·3UF <sub>4</sub>	β-2 2NaF·ZrF <sub>4</sub>
KF·BeF <sub>2</sub>	RbF·6UF <sub>4</sub>	β-3 2NaF·ZrF <sub>4</sub>
KF·2BeF <sub>2</sub>	3RbF·ZrF <sub>4</sub>	β-4 2NaF·ZrF <sub>4</sub>
3KF·CrF <sub>3</sub>	2RbF·ZrF <sub>4</sub>	γ 2NaF·ZrF <sub>4</sub>
KF·LaF <sub>3</sub>	5RbF·4ZrF <sub>4</sub>	7NaF·6ZrF <sub>4</sub>
KF·NaF·BeF <sub>2</sub>	αRbF·ZrF <sub>4</sub>	3NaF·4ZrF <sub>4</sub>
KF·2NaF·UF <sub>4</sub>	βRbF·ZrF <sub>4</sub>	UF <sub>3</sub> ·3ZrF <sub>4</sub>
3KF·3NaF·2ZrF <sub>4</sub>	3RbF·3NaF·4ZrF <sub>4</sub>	ZnF <sub>2</sub> ·ZrF <sub>4</sub>

Beryllium Fluoride,  $\text{BeF}_2$  (1)

$d(\text{\AA})$	$I/I_1$	Lkl
4.09	70	100
3.21	100	101
2.367	100	110
2.189	100	102
2.154	100	111
1.905	70	201
1.748	50	112
1.606	35	202
1.591	20	103
1.550	30	210
1.484	30	211
1.320	30	203
1.233	15	104
1.208	15	302

Cesium Beryllium Fluoride,  $2\text{CsF} \cdot \text{BeF}_2$ 

$d(\text{\AA})$	$I/I_1$
5.34	10
4.46	25
4.02	15
3.80	15
3.62	60
3.40	30
3.28	10
3.23	100
3.08	80
2.871	10
2.685	10
2.636	35
2.543	20
2.398	20
2.355	15
2.242	20
2.164	15
2.106	20
2.079	15
2.025	15
2.000	10
1.959	10
1.901	30
1.822	10
1.805	5
1.770	10
1.724	15
1.673	10
1.620	5
1.584	5
1.541	10

Cesium Beryllium Fluoride,  
CsF·BeF<sub>2</sub>

<u>d(Å)</u>	<u>I/I<sub>1</sub></u>
5.99	10
5.43	50
4.25	5
3.98	10
3.85	35
3.60	30
3.55	35
3.47	85
3.31	10
3.20	100
3.00	40
2.822	10
2.667	5
2.622	10
2.413	10
2.367	25
2.184	10
2.130	20
2.115	10
1.975	10
1.928	5
1.812	15
1.779	10
1.599	10

Cesium Lanthanum Fluoride,  
3CsF·LaF<sub>3</sub>

<u>d(Å)</u>	<u>I/I<sub>1</sub></u>
3.49	100
2.478	35
2.021	60
1.751	20
1.569	12

Cesium Uranium Fluoride  
2CsF·UF<sub>4</sub>

<u>d(Å)</u>	<u>I/I<sub>1</sub></u>
6.92	15
6.19	50
4.90	15
4.04	10
3.98	10
3.93	35
3.80	25
3.65	35
3.59	45
3.55	100
3.44	100
3.31	15
3.29	15
3.09	15
3.00	10
2.93	5
2.89	10
2.814	20
2.731	15
2.652	10
2.475	20
2.453	20
2.286	15
2.226	10
2.101	20
2.060	40
2.034	55
1.955	10
1.815	5
1.776	20
1.770	20
1.708	10
1.640	5
1.601	10

Cesium Uranium Fluoride,  
 $\text{CsF} \cdot \text{UF}_4$ 

$d(\text{\AA})$	$I/I_1$
8.04	45
7.31	70
5.80	20
5.30	10
5.13	5
4.48	10
4.35	5
4.15	20
4.00	80
3.78	15
3.62	100
3.55	70
3.25	20
3.17	10
2.875	40
2.822	10
2.739	5
2.660	20
2.615	5
2.410	35
2.292	10
2.244	10
2.189	10
2.111	5
2.069	35
2.025	25
2.000	30
1.916	10
1.890	5
1.805	60
1.757	5
1.679	5
1.654	10
1.609	10
1.538	5
1.532	5
1.520	5
1.485	5

Cesium Zirconium Fluoride,  
 $2\text{CsF} \cdot \text{ZrF}_4$ 

$d(\text{\AA})$	$I/I_1$
4.11	35
3.88	5
3.73	100
3.54	10
3.34	5
3.20	40
2.96	10
2.771	10
2.687	10
2.529	5
2.501	10
2.430	40
2.286	20
2.140	5
2.056	5
1.932	20
1.857	20
1.779	5
1.773	5
1.667	10
1.609	10
1.601	10
1.480	10

Cesium Zirconium Fluoride  
CsF·ZrF<sub>4</sub>

<u>d(Å)</u>	<u>I/I<sub>1</sub></u>
6.76	35
6.30	15
5.87	10
5.45	10
5.07	15
4.80	10
4.37	20
4.33	20
4.00	15
3.91	25
3.85	15
3.73	100
3.62	85
3.42	50
3.26	85
3.12	40
2.593	15
2.333	10
2.264	15
2.222	25
2.164	10
2.079	15
1.932	15
1.879	15
1.658	10
1.543	15

Chromium(II) Fluoride  
CrF<sub>2</sub>

<u>d(Å)</u>	<u>I/I<sub>1</sub></u>
3.690	20
3.56	5
3.35	50
3.28	25
3.10	10
2.97	100
2.805	30
2.660	5
2.508	15
2.398	10
2.355	10
2.322	5
2.074	10
1.884	15
1.843	10
1.763	15
1.743	10
1.665	10
1.594	10

Chromium(III) Fluoride  
CrF<sub>3</sub>

<u>d(Å)</u>	<u>I/I<sub>1</sub></u>
4.00	30
3.62	100
2.91	5
2.885	5
2.622	25
2.501	10
2.410	10
2.169	25
2.004	5
1.829	10
1.809	20
1.649	30
1.622	5
1.586	10
1.543	10

Chromium Zirconium Fluoride,  
 $\text{CrF}_2 \cdot \text{ZrF}_4$ 

$d(\text{\AA})$	$I/I_1$
4.72	10
4.55	30
4.41	10
4.09	100
4.00	45
3.88	5
2.866	25
2.275	15
2.065	65
1.999	25
1.829	25
1.799	15
1.667	10

Lithium Chromium Fluoride,  
 $3\text{LiF} \cdot \text{CrF}_3$ 

$d(\text{\AA})$	$I/I_1$
4.29	45
4.16	100
3.45	35
2.67	30
2.212	25
2.176	80
2.140	25
2.092	20
2.008	10
1.736	10
1.730	25
1.712	35

Iron Zirconium Fluoride  
 $\text{FeF}_2 \cdot \text{ZrF}_4$ 

$d(\text{\AA})$	$I/I_1$
4.95	10
4.65	40
4.48	45
4.35	8
4.04	100
3.16	15
2.857	45
2.436	8
2.237	15
2.016	50
1.997	25
1.805	65
1.649	25
1.489	15
1.426	10
1.412	15

Lithium Nickel Fluoride  
 $3\text{LiF} \cdot \text{NiF}_2$ 

$d(\text{\AA})$	$I/I_1$
4.81	95
3.06	8
2.508	95
2.074	100
2.016	95
1.599	45
1.470	45
1.426	10
1.405	10

Lithium Uranium Fluoride  
4LiF·UF<sub>4</sub>

<u>d(Å)</u>	<u>I/I<sub>1</sub></u>
5.67	20
5.46	25
5.13	70
4.93	100
4.55	45
4.44	100
4.23	7
3.82	40
3.55	30
3.03	50
2.89	25
2.866	30
2.747	50
2.468	40
2.398	20
2.221	40
2.167	75
2.074	20
2.025	20
1.872	20
1.836	25

Lithium Uranium Fluoride  
3LiF·UF<sub>4</sub> (2)

<u>d(Å)</u>	<u>I/I<sub>1</sub></u>
4.98	20
4.80	15
4.41	100
4.34	100
3.98	15
3.91	8
3.60	80
3.40	10
3.14	25
3.07	50
2.84	80
2.771	30
2.529	35
2.169	15
2.083	75
2.055	35
1.943	50
1.913	25
1.861	30
1.751	25
1.723	25
1.685	25
1.662	8
1.646	20
1.599	8

Lithium Uranium Fluoride  
7LiF·6UF<sub>4</sub>

<u>d(Å)</u>	<u>I/I<sub>1</sub></u>
6.61	6
5.97	20
5.82	15
5.24	90
5.15	10
4.65	10
4.37	13
3.95	55
3.85	13
3.68	20
3.49	75
3.33	90
3.15	70
3.07	10
2.99	95
2.771	30
2.707	30
2.542	25
2.350	13
2.286	25
2.264	13
2.184	10
2.097	30
2.060	30
2.047	75
1.993	25
1.972	20
1.947	25
1.924	15
1.909	30
1.854	45
1.825	20
1.773	20
1.757	25
1.709	15
1.680	15
1.625	15
1.579	25
1.562	8

Lithium Uranium Fluoride  
LiF·4UF<sub>4</sub>

<u>d(Å)</u>	<u>I/I<sub>1</sub></u>
7.02	8
6.33	12
6.07	5
5.73	25
4.98	8
4.70	25
4.25	90
3.88	20
3.78	100
3.52	90
3.16	8
3.13	8
3.06	12
2.84	40
2.771	55
2.542	8
2.350	10
2.310	10
2.226	8
2.000	10
2.088	35
2.016	60
1.991	50
1.888	20
1.819	8
1.767	25



Lithium Zirconium Fluoride  
(low temperature form)  
 $3\text{LiF}\cdot\text{ZrF}_4$

$\overset{\circ}{d}(\text{\AA})$	$I/I_1$
5.49	55
5.40	35
4.88	50
3.67	25
3.43	5
2.91	15
2.79	8
2.67	15
2.40	5
2.07	100
1.94	20
1.82	25
1.80	12
1.78	12
1.65	5
1.59	5
1.57	5

Lithium Zirconium Fluoride  
 $2\text{LiF}\cdot\text{ZrF}_4$

$\overset{\circ}{d}(\text{\AA})$	$I/I_1$
5.14	10
4.75	10
4.62	25
4.29	100
3.15	100
2.75	10
2.49	15
2.42	10
2.38	10
2.26	7
2.19	60
2.15	20
2.05	20
1.95	25
1.70	30
1.63	45
1.58	10
1.54	10

Lithium Zirconium Fluoride  
(high temperature form)  
 $3\text{LiF}\cdot\text{ZrF}_4$

$\overset{\circ}{d}(\text{\AA})$	$I/I_1$
7.2	15
6.42	15
5.68	15
4.58	100
3.75	15
3.68	12
3.43	15
3.24	30
3.15	60
2.84	40
2.63	20
2.535	10
2.361	15
2.20	10
2.047	65
1.847	20

Lithium Zirconium Fluoride  
3LiF·4ZrF<sub>4</sub>

<u>d(Å)</u>	<u>I/I<sub>1</sub></u>
6.11	25
5.24	35
4.90	15
4.21	30
4.00	10
3.90	95
3.77	20
3.69	5
3.33	60
3.29	20
3.26	20
3.16	100
2.615	15
2.303	10
2.248	10
2.227	5
2.194	85
2.159	15
2.043	12
2.130	35
1.947	35
1.912	20
1.883	10
1.721	10

Potassium Beryllium Fluoride  
3KF·BeF<sub>2</sub>

<u>d(Å)</u>	<u>I/I<sub>1</sub></u>
3.42	5
3.30	7
3.07	25
2.98	75
2.81	10
2.64	30
2.54	35
2.51	15
2.39	100
2.27	70
2.04	5
1.96	5
1.85	12
1.78	12
1.73	10
1.72	10
1.69	7
1.60	5
1.56	5
1.54	5

Nickel Zirconium Fluoride  
NiF<sub>2</sub>·ZrF<sub>4</sub>

<u>d(Å)</u>	<u>I/I<sub>1</sub></u>
4.88	15
4.50	5
3.91	100
2.805	50
2.763	10
2.367	15
2.174	5
1.964	50
1.767	55
1.745	15
1.622	10
1.596	10

Potassium Beryllium Fluoride  
 $\text{KF} \cdot \text{BeF}_2$

$\overset{\circ}{d}(\text{\AA})$	$I/I_1$
6.65	15
5.99	60
3.69	5
3.58	10
3.33	90
3.23	35
3.01	100
2.92	5
2.734	15
2.636	20
2.542	5
2.442	5
2.410	5
2.292	15
2.220	7
2.190	5
2.010	35
1.979	25
1.745	7
1.641	10

Potassium Chromium(III) Fluoride  
 $3\text{KF} \cdot \text{CrF}_3$

$\overset{\circ}{d}(\text{\AA})$	$I/I_1$
4.95	35
4.26	15
3.35	35
3.03	100
2.475	25
2.154	20
2.135	55
1.743	15
1.516	15

Potassium Beryllium Fluoride  
 $\text{KF} \cdot 2\text{BeF}_2$

$\overset{\circ}{d}(\text{\AA})$	$I/I_1$
5.99	10
3.66	15
3.31	60
3.00	100
2.65	5
2.55	5
2.44	5
2.39	15
2.37	5
2.29	15
2.21	15
2.14	15
2.08	5
2.01	5
2.00	5
1.83	5
1.78	10
1.60	5

Potassium Lanthanum Fluoride  
 $\text{KF} \cdot \text{LaF}_3$

$\overset{\circ}{d}(\text{\AA})$	$I/I_1$
7.77	100
5.61	70
4.90	35
3.32	40
3.26	60
3.15	65
2.747	100
2.209	40
1.951	25
1.920	20
1.894	12
1.883	30
1.857	65
1.662	20
1.641	5
1.577	20

Potassium Nickel Fluoride<sup>(6)</sup>  
 $2\text{KF}\cdot\text{NiF}_2$

<u>d(Å)</u>	<u>I/I<sub>1</sub></u>
6.51	100
5.85	45
3.83	10
2.94	55
2.830	35
2.344	5
2.212	10
2.173	100
2.135	15
1.999	55
1.909	15
1.805	10
1.730	30
1.654	10
1.633	35
1.594	10
1.474	20

Potassium Sodium Beryllium Fluoride  
 $\text{KF}\cdot\text{NaF}\cdot\text{BeF}_2$

<u>d(Å)</u>	<u>I/I<sub>1</sub></u>
7.08	5
3.85	25
3.17	15
3.08	30
2.866	60
2.786	100
2.675	10
2.626	60
2.593	10
2.528	15
2.410	10
2.367	100
2.287	70
2.212	15
2.189	10
2.159	10
2.125	5
1.995	60
1.967	10
1.819	5
1.806	5
1.777	25
1.691	10
1.620	10
1.605	15

Potassium Sodium Uranium Fluoride  
 $\text{KF}\cdot 2\text{NaF}\cdot\text{UF}_4$

<u>d(Å)</u>	<u>I/I<sub>1</sub></u>
8.64	25
7.80	80
6.00	15
5.42	65
4.93	30
4.44	90
3.51	40
3.46	25
3.16	100
3.13	100
2.90	35
2.830	5
2.710	20
2.593	10
2.557	20
2.462	20
2.443	5
2.344	20
2.321	5
2.222	80
2.189	5
2.156	5
2.074	5
2.047	10
1.995	25
1.977	25
1.947	20
1.872	20
1.833	20
1.809	65
1.776	5
1.754	15
1.727	5
1.654	25
1.606	15
1.579	10
1.560	15

Potassium Sodium Zirconium  
Fluoride,  $3\text{KF} \cdot 3\text{NaF} \cdot 2\text{ZrF}_4$

$d(\text{\AA})$	$I/I_1$
6.11	15
4.90	100
4.82	25
4.55	15
4.27	55
4.09	75
3.78	12
3.40	20
3.28	20
2.99	20
2.97	90
2.72	12
2.682	12
2.585	5
2.455	25
2.417	20
2.368	12
2.270	25
2.226	50
2.135	40
2.051	50
1.931	15
1.822	40
1.748	20
1.721	25

Potassium Sodium Zirconium  
Fluoride,  $3\text{KF} \cdot 2\text{NaF} \cdot 5\text{ZrF}_4$

$d(\text{\AA})$	$I/I_1$
8.51	100
7.70	100
6.20	3
5.57	5
5.15	8
5.01	20
4.77	8
4.65	35
4.29	40
3.86	100
3.75	15
3.50	15
3.40	50
3.23	5
3.17	60
3.09	35
2.87	5
2.80	10
2.58	5
2.47	5
2.39	8
2.33	8
2.19	5
2.16	8
2.14	50
2.09	8
2.06	8
1.95	30
1.94	90
1.91	10
1.89	10
1.88	15
1.84	7
1.81	7
1.76	8
1.70	13

Potassium Sodium Zirconium  
Fluoride,  $\text{KF} \cdot \text{NaF} \cdot \text{ZrF}_4$

$\overset{\text{O}}{d(\text{A})}$	$I/I_1$
5.99	10
5.37	50
4.98	15
4.85	60
4.50	50
4.41	5
4.25	30
4.09	65
3.69	30
3.60	12
3.52	5
3.34	100
3.26	45
3.18	40
3.12	12
3.07	12
3.01	40
2.747	5
2.690	25
2.556	20
2.489	25
2.405	20
2.258	20
2.154	16
2.215	20
2.050	35
1.947	20
1.928	12
1.886	12
1.792	15
1.743	20
1.723	15
1.694	5
1.651	25

Potassium Tellurium Fluoride  
 $\text{KF} \cdot \text{TeF}_4 \cdot \text{H}_2\text{O}$

$d(\text{A})$	$I/I_1$
5.64	100
4.61	60
4.27	100
3.95	10
3.575	75
3.170	15
2.910	25
2.805	50
2.303	25
2.134	10
2.034	10
1.963	45
1.872	20
1.812	10
1.779	10
1.742	15
1.641	5
1.606	10

Potassium Uranium Fluoride (4b)  
7KF·6UF<sub>4</sub>

$\overset{\circ}{d}(\text{Å})$	$I/I_1$
8.11	50
7.50	20
5.54	15
4.93	10
4.82	60
4.46	50
4.35	30
4.17	10
4.06	15
3.78	55
3.58	20
3.44	100
3.29	15
3.12	10
2.97	65
2.885	15
2.858	20
2.710	15
2.550	10
2.515	10
2.409	10
2.368	10
2.333	20
2.321	25
2.298	25
2.243	15
2.200	30
2.140	15
2.111	70
2.096	80
2.047	10
2.030	10
1.995	25
1.987	40
1.983	40
1.959	10
1.913	10
1.896	10
1.861	10
1.840	10
1.802	45
1.793	70
1.789	70
1.757	10
1.730	10

$\overset{\circ}{d}(\text{Å})$	$I/I_1$
1.715	30
1.654	10
1.606	10
1.589	10
1.582	10
1.553	10

Potassium Zirconium Fluoride  
3KF·2ZrF<sub>4</sub>

$\overset{\circ}{d}(\text{Å})$	$I/I_1$
7.37	10
6.56	20
5.95	75
5.54	70
5.09	15
4.95	25
4.90	20
4.39	40
3.97	100
3.83	10
3.71	20
3.66	20
3.56	15
3.45	15
3.33	100
3.31	85
3.11	75
2.571	10
2.550	15
2.270	15
2.214	25
2.200	65
2.111	10
1.979	100
1.883	10
1.842	25
1.739	10
1.705	20
1.673	20

Potassium Zirconium Fluoride  
 $\alpha$ -KF·ZrF<sub>4</sub>

$d(\text{\AA})$	I/I <sub>1</sub>
5.83	60
4.89	35
4.40	80
4.29	20
4.07	5
3.86	100
3.08	20
2.90	10
2.52	5
2.475	15
2.232	5
2.200	20
2.140	15
2.088	10
2.043	5
2.016	10
1.912	5
1.850	10
1.779	5

Potassium Zirconium Fluoride  
 $\beta$ -KF·ZrF<sub>4</sub>

$d(\text{\AA})$	I/I <sub>1</sub>
8.26	7
6.97	50
6.32	100
5.50	15
5.31	30
3.93	15
3.49	40
3.36	35
3.29	60
3.15	100
2.763	7
2.731	10
2.392	15
2.332	15
2.179	12
2.120	12
2.106	40
2.065	15
1.967	10
1.916	60
1.865	15
1.843	35
1.748	15
1.665	5
1.579	70



Rubidium Beryllium Fluoride  
 $3\text{RbF} \cdot \text{BeF}_2$ 

$d(\text{Å})$	$I/I_1$
3.83	10
3.55	10
3.45	35
3.21	10
3.11	10
2.95	60
2.755	10
2.655	5
2.629	5
2.488	55
2.373	5
2.125	25
2.047	15
1.987	5
1.920	10
1.809	15
1.799	15
1.688	40
1.577	35
1.531	5
1.425	100
1.421	100
1.328	10

 Rubidium Beryllium Fluoride  
 $2\text{RbF} \cdot \text{BeF}_2$ 

$d(\text{Å})$	$I/I_1$
5.10	10
4.25	20
3.98	5
3.82	15
3.59	10
3.45	50
3.40	25
3.26	30
3.22	10
3.11	15
3.07	100
2.95	100
2.822	15
2.747	10
2.691	20
2.592	5
2.556	90
2.482	60
2.423	70
2.350	5
2.338	5
2.275	25
2.242	15
2.125	25
2.097	5
2.065	5
2.047	50
1.975	10
1.928	25
1.897	30
1.868	25
1.799	20
1.730	35
1.699	15
1.641	10
1.599	5
1.472	20
1.468	10

Rubidium Beryllium Fluoride  
RbF·BeF<sub>2</sub>

$\overset{\circ}{d}(\text{Å})$	$I/I_1$
6.28	5
5.24	20
4.70	5
4.49	5
4.24	20
4.17	10
3.85	20
3.74	45
3.62	10
3.58	5
3.46	100
3.38	100
3.26	10
3.18	10
3.12	100
3.03	10
2.96	15
2.87	20
2.739	25
2.564	5
2.521	10
2.496	10
2.482	5
2.430	10
2.394	10
2.367	5
2.332	15
2.300	35
2.275	45
2.104	15
2.076	100
2.052	35
1.947	10
1.920	10
1.883	10
1.861	10
1.825	15
1.802	10
1.735	15
1.723	15
1.699	30
1.680	10
1.630	10
1.614	10
1.578	10
1.557	35
1.552	25

Rubidium Beryllium Fluoride  
RbF·2BeF<sub>2</sub>

$\overset{\circ}{d}(\text{Å})$	$I/I_1$
6.16	70
3.39	40
3.31	20
3.10	100
2.715	5
2.475	35
2.410	15
2.327	10
2.201	5
2.149	20
1.876	5
1.852	10
1.819	15
1.552	15

Rubidium Chromium Fluoride  
3RbF·CrF<sub>3</sub>

$\overset{\circ}{d}(\text{Å})$	$I/I_1$
3.47	12
3.16	100
3.13	55
3.11	55
2.98	12
2.57	15
2.468	12
2.350	5
2.315	10
2.264	20
2.208	40
2.149	10
2.088	10
1.829	10
1.813	10
1.805	25
1.625	10
1.582	25
1.582	25

Rubidium Lanthanum Fluoride  
RbF·LaF<sub>3</sub>

$\overset{\circ}{d}(\text{Å})$	$I/I_1$
8.11	25
5.07	15
4.04	8
3.43	100
2.48	25
2.25	40
1.97	15
1.93	15
1.90	40
1.87	12
1.71	15
1.68	15
1.46	8

Rubidium Sodium Uranium Fluoride  
RbF·NaF·UF<sub>4</sub>

$\overset{\circ}{d}(\text{Å})$	$I/I_1$
8.12	100
7.20	10
6.48	30
5.47	60
5.02	10
4.55	55
3.77	45
3.26	80
3.15	60
2.94	10
2.862	5
2.723	10
2.588	15
2.508	20
2.430	15
2.264	60
2.211	5
2.106	10
2.051	10
2.034	75
1.995	10
1.920	10
1.905	30
1.838	40
1.815	15
1.799	10
1.773	7
1.701	25
1.625	25
1.574	5
1.555	10

Rubidium Sodium Beryllium  
Fluoride, RbF·2NaF·BeF<sub>2</sub>

$\overset{\circ}{d}(\text{Å})$	$I/I_1$
4.60	25
4.15	85
3.75	40
2.99	100
2.875	100
2.767	70
2.618	20
2.494	100
2.362	60
2.295	30
2.232	10
2.189	10
2.074	85
1.987	15
1.939	5
1.883	30
1.872	10
1.825	25
1.751	20
1.682	30
1.657	25
1.586	15
1.569	10
1.495	30

Rubidium Sodium Zirconium  
 Fluoride,  $3\text{RbF} \cdot 3\text{NaF} \cdot 2\text{ZrF}_4$  (2)

$d(\text{\AA})$	$I/I_1$
4.32	55
4.15	30
3.57	20
3.33	20
3.22	55
3.09	15
3.00	100
2.747	10
2.481	10
2.450	10
2.281	30
2.164	25
2.125	9
2.079	30
1.920	15
1.868	30
1.767	15
1.742	25
1.500	25

 Rubidium Uranium Fluoride  
 $3\text{RbF} \cdot \text{UF}_4$ 

$d(\text{\AA})$	$I/I_1$
8.43	7
6.03	10
5.56	35
5.13	5
4.83	10
3.80	10
3.72	10
3.58	10
3.48	15
3.40	100
3.24	20
3.01	7
2.89	15
2.593	7
2.521	10
2.482	7
2.398	30
2.275	5
2.194	10
2.140	7
2.101	15
1.959	60
1.912	5
1.897	5
1.844	10
1.694	15
1.623	7

Rubidium Uranium Fluoride  
(low temperature form)  
 $2\text{RbF} \cdot \text{UF}_4$

$\overset{\circ}{d}(\text{Å})$	$I/I_1$
6.03	100
4.75	15
3.49	100
3.23	25
3.17	10
3.02	20
2.814	10
2.367	20
2.281	10
2.184	45
2.008	65
1.963	10
1.947	35
1.909	5
1.740	15
1.699	15
1.673	15
1.635	25
1.584	5
1.538	5
1.511	5
1.476	10

Rubidium Uranium Fluoride  
 $7\text{RbF} \cdot 6\text{UF}_4$

$\overset{\circ}{d}(\text{Å})$	$I/I_1$
8.35	15
4.98	15
4.57	15
4.46	7
4.17	5
3.85	10
3.68	5
3.58	20
3.51	100
3.05	45
2.259	7
2.169	25
2.140	50
2.030	7
1.890	15
1.854	15
1.835	45
1.786	7
1.751	20
1.522	15

Rubidium Uranium Fluoride  
 $\text{RbF} \cdot \text{UF}_4$

$\overset{\circ}{d}(\text{Å})$	$I/I_1$
8.26	5
7.56	10
6.86	100
4.44	15
4.02	15
3.96	10
3.85	5
3.79	5
3.56	10
3.46	50
3.43	100
3.25	5
3.04	12
2.630	5
2.607	10
2.527	10
2.315	5
2.286	20
2.201	10
2.164	7
2.130	7
2.008	5
1.987	30
1.916	5
1.901	10
1.865	5
1.850	5
1.829	5
1.733	5
1.714	25
1.579	5
1.518	5

Rubidium Uranium Fluoride $2\text{RbF} \cdot 3\text{UF}_4$		<sup>o</sup> <u>d(A)</u>	<u>I/I<sub>1</sub></u>
6.39	12	3.59	15
5.80	45	3.52	100
5.04	7	3.43	10
4.18	7	3.36	80
3.86	40	3.34	35
3.50	100	3.24	60
3.22	15	3.10	35
3.07	5	3.00	5
2.780	10	2.98	5
2.614	15	2.90	40
2.417	10	2.848	10
2.361	10	2.736	10
2.259	25	2.706	20
2.184	30	2.667	10
2.038	60	2.629	20
1.943	10	2.571	15
1.935	5	2.501	10
1.843	10	2.469	10
1.805	10	2.433	5
1.773	7	2.404	15
		2.341	15
		2.315	5
		2.247	15
		2.237	15
		2.216	40
		2.159	20
		2.111	5
		2.088	20
		2.070	10
		2.043	15
		2.034	20
		2.025	80
		2.002	100
		1.951	60
		1.903	10
		1.883	10
		1.870	40
		1.847	15
		1.831	30
		1.809	10
		1.789	10
		1.757	25
		1.720	35
		1.659	25
		1.614	10
		1.594	20
		1.589	15
		1.577	5
		1.564	10
		1.550	20

Rubidium Uranium Fluoride $\text{RbF} \cdot 3\text{UF}_4$		<sup>o</sup> <u>d(A)</u>	<u>I/I<sub>1</sub></u>
7.97	60	2.070	10
7.63	10	2.043	15
7.31	15	2.034	20
6.71	10	2.025	80
6.32	5	2.002	100
5.80	5	1.951	60
5.57	10	1.903	10
5.50	20	1.883	10
5.38	10	1.870	40
5.01	10	1.847	15
4.55	30	1.831	30
4.41	15	1.809	10
4.31	5	1.789	10
4.23	25	1.757	25
4.15	15	1.720	35
4.09	5	1.659	25
4.04	10	1.614	10
3.99	20	1.594	20
3.88	25	1.589	15
3.73	40	1.577	5
3.66	55	1.564	10
		1.550	20

Rubidium Uranium Fluoride  
RbF·6UF<sub>4</sub>

$d(\text{Å})$	$I/I_1$
7.14	15
6.53	20
4.82	10
4.56	25
4.35	30
4.13	75
3.93	20
3.85	25
3.68	10
3.55	50
3.47	100
3.30	10
3.22	10
3.11	10
3.05	5
2.98	60
2.90	25
2.747	10
2.686	20
2.410	15
2.367	5
2.270	15
2.172	20
2.049	75
2.030	10
2.006	20
1.957	45
1.833	40
1.763	20
1.724	5
1.688	15
1.641	10
1.630	10
1.620	5
1.606	5

Rubidium Zirconium Fluoride  
3RbF·ZrF<sub>4</sub>

simple cubic  $a_0 = 3.288 \text{ Å}$

$d(\text{Å})$	$I/I_1$
3.290	100
2.326	60
1.900	100
1.646	20

Rubidium Zirconium Fluoride  
2RbF·ZrF<sub>4</sub>  
(low temperature form)

5.350	10
4.835	8
3.590	100
3.080	34
2.414	5
2.338	40
2.199	20
1.861	15
1.789	15
1.569	12
1.422	8

Rubidium Zirconium Fluoride  
5RbF·4ZrF<sub>4</sub>

<u>d(Å)</u>	<u>I/I<sub>1</sub></u>
5.87	10
5.61	10
5.04	10
4.77	15
4.50	12
4.29	65
3.98	25
3.82	30
3.78	15
3.72	15
3.63	30
3.59	95
3.52	8
3.44	100
3.42	100
3.36	100
3.29	100
3.05	12
2.60	25
2.508	30
2.386	12
2.350	75
2.281	10
2.275	12
2.260	12
2.253	12
2.245	8
2.226	12
2.152	45
2.067	12
2.027	12
1.959	30
1.932	30
1.897	15
1.868	20
1.833	20
1.770	30
1.763	30
1.679	12

Rubidium Zirconium Fluoride  
α-RbF·ZrF<sub>4</sub>

<u>d(Å)</u>	<u>I/I<sub>1</sub></u>
7.13	40
5.95	20
4.28	10
4.02	10
3.83	80
3.69	25
3.59	5
3.45	100
3.42	100
3.34	100
3.07	15
3.01	10
2.699	10
2.622	10
2.574	20
2.550	7
2.450	10
2.423	25
2.327	14
2.226	10
2.184	10
2.149	10
2.118	10
2.056	25
1.991	10
1.957	25
1.928	10
1.905	20
1.868	15
1.812	15
1.782	10
1.763	10
1.742	10
1.670	15
1.611	10
1.547	10
1.502	15



Rubidium Zirconium Fluoride  
 $\beta$ -RbF·ZrF<sub>4</sub>

$d(\text{\AA})$	$I/I_1$	$d(\text{\AA})$	$I/I_1$
7.37	12	2.191	30
6.66	40	2.116	40
6.32	5	2.074	10
5.53	5	2.034	5
5.17	5	1.997	30
5.01	20	1.989	50
4.67	5	1.935	50
4.54	5	1.929	50
4.46	5	1.926	55
4.35	12	1.903	30
3.99	12	1.894	5
3.88	5	1.872	12
3.77	30	1.857	10
3.72	35	1.836	10
3.69	40	1.773	5
3.61	40	1.757	5
3.42	100	1.751	5
3.36	100	1.742	5
3.325	100	1.731	15
3.090	10	1.728	15
3.060	15	1.719	24
2.960	5	1.716	15
2.900	5	1.709	15
2.822	5	1.699	10
2.788	40	1.694	10
2.767	45	1.679	20
2.678	5	1.662	5
2.614	5	1.654	20
2.564	5		
2.514	8		
2.501	8		
2.482	5		
2.455	5		
2.433	5		
2.423	5		
2.398	5		
2.380	10		
2.355	10		
2.344	10		
2.309	5		
2.292	8		
2.253	5		
2.240	12		
2.211	15		
2.206	15		

Rubidium Sodium Zirconium  
 Fluoride,  $3\text{RbF}\cdot 3\text{NaF}\cdot 4\text{ZrF}_4$ 

$d(\text{\AA})$	$I/I_1$
6.77	5
6.11	15
5.72	15
5.37	30
4.67	9
4.21	35
4.11	10
3.98	15
3.82	20
3.72	30
3.56	5
3.50	10
3.44	100
3.31	8
3.22	30
3.17	65
3.11	100
3.07	5
3.01	20
2.95	5
2.571	10
2.482	8
2.356	8
2.243	8
2.164	10
2.111	15
2.088	15
2.060	35
2.025	30
2.004	8
1.976	8
1.947	30
1.909	20
1.861	85
1.812	5
1.727	10
1.638	10
1.591	10

 Rubidium Sodium Zirconium  
 Fluoride,  $\text{RbF}\cdot \text{NaF}\cdot 2\text{ZrF}_4$ 

$d(\text{\AA})$	$I/I_1$
7.43	5
6.71	5
6.54	5
5.91	40
4.55	20
4.41	5
4.13	50
3.90	8
3.71	60
3.52	45
3.40	8
3.35	100
3.29	12
3.23	10
3.13	12
2.826	20
2.481	5
2.430	5
2.392	5
2.380	5
2.344	5
2.286	5
2.237	5
2.194	10
2.159	20
2.142	10
2.083	5
2.060	10
1.983	8
1.963	12
1.943	15
1.935	25
1.920	5
1.854	8
1.836	8
1.757	40
1.754	45
1.694	8
1.676	40
1.670	30

Rubidium Sodium Fluoride  
RbF · NaF · ZrF<sub>4</sub>

$d(\text{Å})$	$I/I_1$
5.21	10
4.82	5
4.68	40
4.21	45
3.75	15
3.65	5
3.59	10
3.39	65
3.34	10
3.29	20
3.24	35
2.571	10
2.514	8
2.350	15
2.332	30
2.321	30
2.152	100
2.094	100
1.886	10
1.763	20
1.614	8

Sodium Cerium Fluoride  
NaF · CeF<sub>3</sub>

$d(\text{Å})$	$I/I_1$
5.95	15
5.35	60
3.43	40
3.09	90
2.57	100
2.411	10
2.392	15
2.179	35
2.097	5
2.016	5
1.967	25
1.894	10
1.843	10
1.815	10
1.777	90
1.612	10
1.538	15
1.375	15

Rubidium Uranium Fluoride  
RbF · UF<sub>3</sub>

$d(\text{Å})$	$I/I_1$
8.12	30
5.07	15
4.04	7
3.43	100
2.485	25
2.253	40
1.968	15
1.932	12
1.898	40
1.869	40
1.715	15
1.682	15
1.460	8

Sodium Chromium (II) Fluoride  
NaF · CrF<sub>2</sub>

$d(\text{Å})$	$I/I_1$
5.15	5
4.63	15
4.49	10
3.95	15
3.01	100
2.593	10
2.368	10
2.189	15
2.039	25
1.975	20

Sodium Hafnium Fluoride  
3NaF·HfF<sub>4</sub>

<u>d(Å)</u>	<u>I/I<sub>1</sub></u>
5.27	55
4.27	90
3.77	50
3.07	100
2.94	40
1.967	30
1.883	10
1.872	75
1.770	15
1.603	50
1.586	5

Sodium Hafnium Fluoride  
NaF·HfF<sub>4</sub>

<u>d(Å)</u>	<u>I/I<sub>1</sub></u>
7.37	30
5.61	100
5.37	80
5.04	15
4.37	30
4.17	20
4.05	45
3.98	60
3.87	100
3.69	10
3.45	15
3.35	20
3.31	15
3.26	15
3.13	85
3.03	20
2.85	20
2.82	20
2.75	15
2.71	40
2.62	10
2.51	35
2.49	15
2.28	40
2.21	40
2.03	10
2.02	10
1.99	15
1.95	15
1.91	60
1.86	35
1.84	10
1.81	10
1.78	35
1.73	15
1.71	40
1.66	25
1.65	55
1.63	65
1.61	5
1.58	10
1.56	15
1.55	15

Sodium Iron (II) Fluoride  
NaF·FeF<sub>2</sub>

<u>d(Å)</u>	<u>I/I<sub>1</sub></u>
4.35	25
3.92	90
3.08	15
2.83	30
2.78	50
2.731	12
2.462	15
2.174	45
1.964	100
1.783	35
1.757	20
1.739	12

Sodium Lithium Chromium Fluoride  
NaF·2LiF·CrF<sub>3</sub>

<u>d(Å)</u>	<u>I/I<sub>1</sub></u>
4.36	100
4.17	20
3.17	10
3.08	60
2.76	15
2.68	15
2.629	15
2.515	20
2.416	20
2.327	10
2.247	30
2.216	10
2.176	20
2.097	10
2.012	5
1.995	30
1.947	40
1.708	20

Sodium Lanthanum Fluoride (4a)  
NaF·LaF<sub>3</sub>

<u>d(Å)</u>	<u>I/I<sub>1</sub></u>
5.37	65
3.11	100
2.09	100
2.69	15
2.405	20
2.194	90
2.021	15
1.913	15
1.783	80
1.627	15
1.545	12
1.485	10
1.383	20

Sodium Nickel Fluoride  
NaF·NiF<sub>2</sub>

<u>d(Å)</u>	<u>I/I<sub>1</sub></u>
4.25	10
3.85	100
3.45	8
3.33	5
3.05	5
3.01	10
2.764	35
2.723	50
2.682	20
2.398	10
2.327	10
2.309	25
2.130	15
2.070	10
2.034	25
1.924	100
1.796	5
1.757	20
1.742	35
1.720	30
1.697	10
1.586	20
1.579	20
1.557	40

Sodium Thorium Fluoride (4a)  
2NaF·ThF<sub>4</sub>

<u>d(Å)</u>	<u>I/I<sub>1</sub></u>
6.22	10
5.72	45
5.38	10
5.17	100
4.53	10
4.11	20
3.83	5
3.69	5
3.65	5
3.40	10
3.31	15
3.07	35
2.99	70
2.863	20
2.590	55
2.442	5
2.374	10
2.352	20
2.197	40
2.056	5
1.987	5
1.955	15
1.928	15
1.909	10
1.793	10
1.742	55
1.724	20
1.705	5
1.609	5
1.572	5

Sodium Uranium (III) Fluoride  
 $\text{NaF} \cdot \text{UF}_3$

$\overset{\circ}{d}(\text{Å})$	$I/I_1$
5.34	30
3.09	100
2.392	15
2.179	55
2.016	50
1.898	10
1.779	75
1.614	10
1.541	10

Sodium Uranium Fluoride (3)  
 $\alpha \text{ 3NaF} \cdot \text{UF}_4$

$\overset{\circ}{d}(\text{Å})$	$I/I_1$
5.47	35
4.87	100
3.85	50
3.15	75
3.03	30
2.721	30
2.435	30
2.378	25
2.224	30
2.022	15
1.926	45
1.815	25
1.791	30
1.722	35
1.642	30
1.574	15
1.511	15
1.496	15
1.457	15

Sodium Uranium Fluoride (3)  
 $\beta \text{ 3NaF} \cdot \text{UF}_4$

$\overset{\circ}{d}(\text{Å})$	$I/I_1$
6.28	45
5.15	100
3.98	60
3.65	10
3.29	10
2.97	100
2.578	25
2.206	10
2.164	30
2.102	80
1.991	20
1.943	15
1.897	5
1.857	15
1.819	10
1.751	10
1.718	85
1.552	25
1.506	10
1.485	15

Sodium Uranium Fluoride (2,3,4a)  
 $\beta \text{ 2-2NaF} \cdot \text{UF}_4$

$\overset{\circ}{d}(\text{Å})$	$I/I_1$
5.12	100
3.72	10
3.02	85
2.97	40
2.578	35
2.327	55
2.121	90
1.947	35
1.869	15
1.754	15
1.730	100
1.583	15
1.562	25

Sodium Uranium Fluoride (3)  
 $\beta$ -3  $2\text{NaF}\cdot\text{UF}_4$

<u>d(Å)</u>	<u>I/I<sub>1</sub></u>	<u>hkl</u>
7.25	100	001
5.31	20	100
4.28	75	101
3.06	35	110
2.99	60	102
2.817	10	111
2.644	5	200
2.488	15	201
2.410	20	003
2.338	5	112
2.194	10	103
2.135	35	202
1.999	5	210
1.928	15	211
1.890	15	113
1.809	20	004
1.783	5	203
1.760	15	300
1.748	25	212
1.712	15	104
1.555	10	114
1.538	5	213
1.527	5	220
1.491	5	221, 204
1.436	5	311
1.422	5	303
1.394	15	105
1.358	5	312
1.340	5	214

Sodium Uranium Fluoride (2,3)  
 $\gamma$   $2\text{NaF}\cdot\text{UF}_4$

<u>d(Å)</u>	<u>I/I<sub>1</sub></u>
5.79	50
4.92	100
3.78	40
3.24	50
3.17	70
2.91	20
2.831	40
2.767	30
2.501	20
2.235	50
2.173	50
2.151	50
2.003	50
1.967	100
1.947	20
1.896	20
1.860	30
1.826	20
1.696	10
1.669	80



Sodium Uranium Fluoride (4c,5)  
5NaF·3UF<sub>4</sub>

$d(\text{Å})$	$I/I_1$
3.19	100
2.777	30
1.964	65
1.684	65
1.609	15
1.396	15
1.282	-
1.248	-
1.144	-

Sodium Uranium Fluoride (4d,5)  
7NaF·6UF<sub>4</sub>

$d(\text{Å})$	$I/I_1$
7.76	40
5.34	20
4.55	50
4.32	70
4.25	55
3.87	10
3.68	25
3.43	10
3.33	85
3.25	100
3.03	20
2.97	20
2.863	50
2.786	50
2.595	15
2.585	30
2.437	10
2.342	10
2.177	35
2.114	25
2.073	10
2.047	10
2.041	30
1.909	15
1.731	40
1.773	45
1.664	10

Sodium Uranium Fluoride  
NaF·2UF<sub>4</sub>

$d(\text{Å})$	$I/I_1$	$d(\text{Å})$	$I/I_1$
7.80	8	2.449	8
6.61	8	2.411	8
6.19	20	2.355	10
6.03	20	2.290	10
5.61	100	2.250	8
5.47	25	2.211	8
5.21	12	2.184	8
4.98	10	2.152	8
4.92	10	2.144	8
4.80	8	2.137	8
4.65	25	2.128	25
4.33	35	2.090	8
4.09	10	2.065	30
4.04	15	2.034	15
3.91	15	2.010	10
3.80	8	1.983	15
3.69	50	1.947	15
3.60	15	1.939	20
3.51	15	1.931	30
3.43	20	1.912	35
3.42	20	1.865	10
3.41	20	1.840	8
3.30	45	1.817	25
3.26	70	1.792	8
3.15	20	1.790	8
3.12	8	1.754	10
3.08	65	1.742	10
2.98	8	1.730	20
2.90	8	1.720	15
2.87	10	1.643	15
2.813	45	1.638	10
2.747	8	1.586	15
2.731	8	1.584	10
2.675	8	1.576	15
2.622	8	1.572	15
2.600	8	1.554	20

Sodium Zirconium Fluoride  
 $3\text{NaF} \cdot \text{ZrF}_4$  (3)

$d(\text{Å})$	$I/I_1$
5.25	35
4.75	75
3.75	20
3.06	100
2.93	10
2.32	10
1.97	5
1.88	20
1.87	35
1.77	5
1.60	15
1.59	5
1.46	5

Sodium Zirconium Fluoride  
 $\beta$ -2  $2\text{NaF} \cdot \text{ZrF}_4$  (3)

$d(\text{Å})$	$I/I_1$
5.72	20
5.47	80
5.38	50
3.78	80
3.31	20
3.11	100
2.675	20
2.410	20
2.216	20
2.043	20
1.912	100
1.897	80
1.793	20
1.645	20
1.617	50
1.557	10

Sodium Zirconium Fluoride  
 $\beta$ -1  $2\text{NaF} \cdot \text{ZrF}_4$  (3)

$d(\text{Å})$	$I/I_1$
5.45	10
5.15	55
4.73	25
4.06	15
3.83	5
3.72	40
3.48	15
3.43	5
3.15	5
3.09	100
2.839	5
2.682	10
2.593	5
2.327	10
2.237	5
2.206	10
2.169	5
2.088	10
2.038	15
1.995	10
1.951	6
1.890	100
1.847	5
1.812	5
1.776	10
1.711	5
1.688	5
1.630	5
1.611	45

Sodium Zirconium Fluoride  
 $\beta$ -3  $2\text{NaF} \cdot \text{ZrF}_4$  (3)

$d(\text{Å})$	$I/I_1$
5.12	100
4.80	10
4.55	100
4.00	5
3.89	10
3.60	90
3.24	25
3.12	10
3.04	10
2.95	40
2.893	100
2.557	10
2.550	10
2.475	40
2.275	30
2.221	30
2.085	10
1.894	100
1.799	30
1.708	60
1.608	6
1.543	20

## Sodium Zirconium Fluoride

 $\beta$ -4  $2\text{NaF} \cdot \text{ZrF}_4$ 

$d(\text{\AA})$	$I/I_1$
7.97	35
5.54	15
5.12	80
4.88	65
4.48	15
4.33	20
4.00	70
3.83	70
3.25	100
3.12	50
3.04	40
2.89	15
2.759	25
2.666	20
2.564	20
2.398	15
2.361	15
2.327	15
2.034	20
1.979	15
1.935	20
1.912	50
1.796	15
1.751	10
1.711	20
1.619	10
1.596	20
1.574	15

## Sodium Zirconium Fluoride

 $\gamma$   $2\text{NaF} \cdot \text{ZrF}_4$ 

$d(\text{\AA})$	$I/I_1$
4.98	100
4.37	50
4.19	80
3.96	50
3.45	50
3.36	60
3.16	40
3.07	80
2.839	70
1.843	50

## Sodium Zirconium Fluoride

 $7\text{NaF} \cdot 6\text{ZrF}_4$  (5)

$d(\text{\AA})$	$I/I_1$
7.37	20
6.86	10
5.07	10
4.85	5
4.37	20
4.06	23
3.98	5
3.69	10
3.55	5
3.45	40
3.25	5
3.13	100
2.85	5
2.72	15
2.64	5
2.37	5
2.32	5
2.19	5
2.09	10
2.03	10
2.00	15
1.95	15
1.92	100
1.91	100
1.86	15
1.81	10
1.79	10
1.74	10
1.70	10
1.68	10
1.63	50
1.61	5
1.58	5
1.56	10
1.51	15
1.48	10

Sodium Zirconium Fluoride  
NaF·ZrF<sub>4</sub> (2)

<u>d(Å)</u>	<u>I/I<sub>1</sub></u>
5.57	30
3.37	40
3.96	30
3.86	100
3.01	15
2.86	10
2.75	7
2.72	7
2.51	7
2.28	25
2.21	20
2.09	30
2.05	5
2.02	15
2.00	5
1.99	10
1.86	15
1.84	7
1.70	20
1.65	20
1.63	12
1.61	7
1.56	10

Sodium Zirconium Fluoride  
3NaF·4ZrF<sub>4</sub> (5)

<u>d(Å)</u>	<u>I/I<sub>1</sub></u>
7.56	15
7.42	45
5.47	15
4.15	100
3.78	20
3.74	20
3.42	25
3.39	35
3.36	60
2.630	15
2.074	75
1.935	12
1.766	50
1.506	25

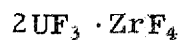
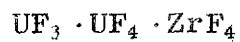
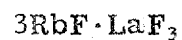
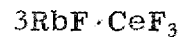
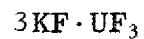
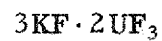
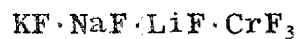
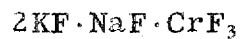
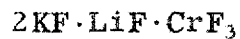
Uranium Zirconium Fluoride  
UF<sub>3</sub>·3ZrF<sub>4</sub>

<u>d(Å)</u>	<u>I/I<sub>1</sub></u>
6.15	35
4.11	100
3.95	20
3.88	8
3.72	75
3.44	35
3.22	20
3.17	35
2.77	12
2.69	30
2.51	25
2.29	20
2.25	17
2.12	12
2.06	85
2.02	10
2.01	10
1.98	8
1.95	10
1.92	35
1.86	30
1.84	35
1.81	20
1.77	20
1.73	20
1.68	15
1.64	5
1.62	7
1.59	7
1.58	7
1.54	10

Zinc Zirconium Fluoride  
 $\text{ZnF}_2 \cdot \text{ZrF}_4$

$d(\text{\AA})$	$I/I_1$
4.68	5
4.49	35
4.15	5
4.06	100
3.55	15
3.15	15
2.852	50
2.226	15
2.008	35
1.989	20
1.815	10
1.796	65
1.793	50
1.638	30

B. X-ray diffraction patterns for the compounds below are listed in this section. The purity of these compounds has never been absolutely established, nor have these compounds been isolated as single phases.



Potassium Lithium Chromium (III)  
Fluoride  $2\text{KF} \cdot \text{LiF} \cdot \text{CrF}_3$

$\overset{\circ}{d}(\text{Å})$	$I/I_1$
4.61	75
3.33	20
2.95	15
2.822	90
2.667	15
2.482	15
2.407	30
2.327	30
2.303	90
2.012	45
1.993	100
1.672	15
1.628	40
1.536	15

Potassium Sodium Lithium Chromium (III) Fluoride  
 $\text{KF} \cdot \text{NaF} \cdot \text{LiF} \cdot \text{CrF}_3$

$\overset{\circ}{d}(\text{Å})$	$I/I_1$
4.74	45
4.09	10
2.90	100
2.478	15
2.373	55
2.321	20
2.052	85
2.010	30
1.676	40
1.582	10

Potassium Sodium Chromium (III)  
Fluoride  $2\text{KF} \cdot \text{NaF} \cdot \text{CrF}_3$

$\overset{\circ}{d}(\text{Å})$	$I/I_1$
4.78	30
4.14	20
2.93	95
2.495	10
2.385	65
2.292	5
2.069	100
1.688	55

Potassium Uranium Fluoride  
 $3\text{KF} \cdot 2\text{UF}_3$

$\overset{\circ}{d}(\text{Å})$	$I/I_1$
5.37	60
4.67	25
3.43	90
3.25	100
2.97	60
2.79	20
2.73	20
2.33	20
2.10	80
1.98	40
1.79	75
1.71	20
1.53	20
1.36	35



Potassium Uranium Fluoride  
 $3\text{KF} \cdot \text{UF}_3$

$d(\text{Å})$	$I/I_1$
6.27	70
5.68	30
5.24	100
3.69	70
3.47	40
3.24	50
2.88	20
2.60	80
2.50	20
2.164	30
2.074	15
1.955	20
1.928	30
1.840	25
1.730	20

Rubidium Lanthanum Fluoride  
 $3\text{RbF} \cdot \text{LaF}_3$

$d(\text{Å})$	$I/I_1$
4.93	30
3.66	30
3.47	20
3.40	20
3.30	90
3.11	60
2.821	100
2.344	40
2.237	20
2.206	15
2.184	20
1.991	40
1.905	30
1.701	40
1.627	20

Rubidium Cerium Fluoride  
 $3\text{RbF} \cdot \text{CeF}_3$

$d(\text{Å})$	$I/I_1$
5.91	15
4.93	15
3.68	25
3.58	20
3.45	45
3.31	100
3.24	55
3.21	35
3.05	15
2.608	15
2.522	40
2.355	45
2.296	20
2.088	10
2.060	10
1.968	10
1.943	10
1.924	25
1.905	15
1.882	25
1.840	15
1.812	15
1.789	10
1.694	15

Uranium Zirconium Fluoride  
 $UF_3 \cdot UF_4 \cdot ZrF_4$

<u>d(Å)</u>	<u>I/I<sub>1</sub></u>
6.42	15
4.07	100
3.98	65
3.58	12
3.39	40
3.22	30
3.01	40
2.63	20
2.08	50
2.02	12
1.98	15
1.94	12
1.89	50
1.80	25
1.54	20

Uranium Zirconium Fluoride  
 $2UF_3 \cdot ZrF_4$

<u>d(Å)</u>	<u>I/I<sub>1</sub></u>
4.05	100
3.98	25
3.58	15
3.39	40
3.22	35
3.00	40
2.63	15
2.07	40
2.02	12
1.94	30
1.89	40
1.84	12
1.79	25

Footnotes

1. This compound is rhombohedral and is isostructural with  $\text{SiO}_2$ .
2. This compound has only metastable existence in a system of its own composition.
3. The original measurements on this material were made by P. A. Agron (Chemistry Division), B. S. Borie, Jr., and R. M. Steele (Metallurgy Division).
4.
  - a. This compound has been reported by W. H. Zachariasen, J. Am. Chem. Soc. 70, 2147-51 (1948)
  - b. This compound was reported by W. H. Zachariasen, idem., to be  $\text{KF} \cdot \text{UF}_4$ .
  - c. This compound was reported by W. H. Zachariasen, idem., to be  $\alpha\text{-2NaF} \cdot \text{UF}_4$ .
  - d. This compound was reported by W. H. Zachariasen, idem., to be  $\text{NaF} \cdot \text{UF}_4$ .
5. The original ORNL measurements on this material were made by P. A. Agron (Chemistry Division), B. S. Borie, Jr., and R. M. Steele (Metallurgy Division). The compound formula was established at a later date by the authors.
6. A graphical representation of a Debye-Scherrer pattern for this compound is reported by Wagner and Balz, Z. Electrochem. 56, 576 (1952).



Compound	dA°			I/I <sub>1</sub>			Page
3KF·2NaF·5ZrF <sub>4</sub>	8.51	7.70	3.86	100	100	100	35
RbF·NaF·UF <sub>4</sub>	8.12	3.26	2.034	100	80	75	41
KF·LaF <sub>3</sub>	7.77	2.747	5.61	100	100	70	33
B-32NaF·UF <sub>4</sub>	7.25	4.28	2.99	100	75	60	54
RbF·UF <sub>4</sub>	6.86	3.43	3.46	100	100	50	43
2KF·NiF <sub>2</sub>	6.51	2.173	1.999	100	100	55	34
B·KF·ZrF <sub>4</sub>	6.32	3.15	1.579	100	100	70	38
2RbF·UF <sub>4</sub>	6.03	3.49	1.963	100	100	65	43
KF·TeF <sub>4</sub> ·H <sub>2</sub> O	5.64	4.27	3.575	100	100	75	36
NaF·HfF <sub>4</sub>	5.61	3.87	3.13	100	100	85	50
NaF·2UF <sub>4</sub>	5.61	3.26	3.08	100	70	65	56
2NaF·THF <sub>4</sub>	5.17	2.99	2.590	100	70	55	52
B3NaF·UF <sub>4</sub>	5.15	2.97	1.718	100	100	85	53
B-32NaFZrF <sub>4</sub> (3)	5.12	4.55	2.893	100	100	100	57
B-22NaF·UF <sub>4</sub>	5.12	1.730	2.121	100	100	90	53
γ2NaF·ZrF <sub>4</sub>	4.98	4.19	3.07	100	100	80	58
4LiF·UF <sub>4</sub>	4.93	4.44	5.13	100	100	70	29
γ-2NaF·UF <sub>4</sub>	4.92	1.967	1.669	100	100	80	54
3KF·3NaF·2ZrF <sub>4</sub>	4.90	2.72	3.78	100	90	75	35
α-3NaF·UF <sub>4</sub>	4.87	3.15	3.85	100	75	50	53
3LiF·UF <sub>4</sub>	4.41	4.43	3.60	100	100	80	29
NaF·2LiF·CrF <sub>3</sub>	4.36	3.08	1.947	100	60	40	51
2LiF·ZrF <sub>4</sub>	4.29	3.15	2.19	100	100	60	31
3LiF·CrF <sub>3</sub>	4.16	2.176	4.29	100	80	45	28
3NaF·4ZrF <sub>4</sub> (5)	4.15	2.074	2.630	100	75	60	59

1  
 2  
 3  
 4  
 5  
 6  
 7  
 8  
 9  
 10  
 11  
 12  
 13  
 14  
 15  
 16  
 17  
 18  
 19  
 20  
 21  
 22  
 23  
 24  
 25  
 26  
 27  
 28  
 29  
 30  
 31  
 32  
 33  
 34  
 35  
 36  
 37  
 38  
 39  
 40  
 41  
 42  
 43  
 44  
 45  
 46  
 47  
 48  
 49  
 50  
 51  
 52  
 53  
 54  
 55  
 56  
 57  
 58  
 59  
 60  
 61  
 62  
 63  
 64  
 65  
 66  
 67  
 68  
 69  
 70  
 71  
 72  
 73  
 74  
 75  
 76  
 77  
 78  
 79  
 80  
 81  
 82  
 83  
 84  
 85  
 86  
 87  
 88  
 89  
 90  
 91  
 92  
 93  
 94  
 95  
 96  
 97  
 98  
 99  
 100

Compound	$dA^{\circ}$			I/I <sub>1</sub>			Page
UF <sub>3</sub> ·2ZrF <sub>4</sub>	4.11	2.06	3.72	100	85	75	59
CrF <sub>2</sub> ·ZrF <sub>4</sub>	4.09	2.065	4.00	100	65	45	28
ZnF <sub>2</sub> ·ZrF <sub>4</sub>	4.06	1.796	2.852	100	65	50	60
FeF <sub>2</sub> ·ZrF <sub>4</sub>	4.04	1.805	2.016	100	65	50	28
3KF·2ZrF <sub>4</sub>	3.97	3.33	1.979	100	100	100	37
NiF <sub>2</sub> ·ZrF <sub>4</sub>	3.91	1.767	2.805	100	55	50	32
α-KF·ZrF <sub>4</sub>	3.86	4.40	5.83	100	80	60	38
NaF·ZrF <sub>4</sub> (2)	3.86	3.37	5.57	100	40	30	59
NaF·NiF <sub>2</sub>	3.85	1.924	2.723	100	100	50	52
LiF·4UF <sub>4</sub>	3.78	4.25	3.52	100	90	90	30
CsF·ZrF <sub>4</sub>	3.73	3.62	3.26	100	85	85	27
2CsF·ZrF <sub>4</sub>	3.73	3.20	2.43	100	40	40	26
CrF <sub>3</sub>	3.62	4.00	1.649	100	30	30	27
2RbF·ZrF <sub>4</sub>	3.590	2.338	3.080	100	40	34	45
2CsF·UF <sub>4</sub>	3.55	3.44	2.03	100	100	55	25
RbF·3UF <sub>4</sub>	3.52	2.002	3.36	100	100	80	44
7RbF·6UF <sub>4</sub>	3.51	2.140	3.05	100	50	45	43
2RbF·3UF <sub>4</sub>	3.50	2.038	5.80	100	60	45	44
3CsF·LaF <sub>3</sub>	3.49	3.62	4.00	100	100	80	25
RbF6UF <sub>4</sub>	3.47	3.93	2.030	100	75	75	45
RbF·BeF <sub>2</sub>	3.46	3.38	3.12	100	100	100	40
α·RbF·ZrF <sub>4</sub>	3.45	3.42	3.34	100	100	100	46
5RbF·4ZrF <sub>4</sub>	3.44	3.42	3.36	100	100	100	46
3RbF·3NaF·4ZrF <sub>4</sub>	3.44	3.11	1.861	100	100	85	48
7KF·6UF <sub>4</sub>	3.44	2.096	2.111	100	80	70	37
RbF·UF <sub>3</sub>	3.43	2.253	1.898	100	40	40	49
RbF·LaF <sub>3</sub>	3.43	2.25	1.87	100	40	40	41





Compound	$dA^\circ$			I/I <sub>1</sub>			Page
B-RbF-ZrF <sub>4</sub>	3.42	3.36	3.325	100	100	100	47
3RbF·UF <sub>4</sub>	3.40	1.959	5.56	100	60	35	42
RbF·NaF·2ZrF <sub>4</sub>	3.35	3.71	4.13	100	60	50	48
KF·NaF·ZrF <sub>4</sub>	3.34	4.09	4.85	100	65	60	36
3RbF·ZrF <sub>4</sub>	3.290	1.900	2.326	100	100	60	45
B4 2NaF·ZrF <sub>4</sub>	3.25	5.12	4.00	100	80	70	58
7NaF·6UF <sub>4</sub>	3.25	3.33	4.32	100	85	70	55
2CsF·BeF <sub>2</sub>	3.23	3.08	3.62	100	80	60	24
BeF <sub>2</sub>	3.21	2.367	2.189	100	100	100	24
CsF·BeF <sub>2</sub>	3.20	5.43	3.00	100	50	40	25
5NaF·3UF <sub>4</sub>	3.19	1.964	1.684	100	65	65	55
3LiF·4ZrF <sub>4</sub>	3.16	3.90	2.194	100	95	80	32
KF·2NaF·UF <sub>4</sub>	3.16	3.13	4.44	100	100	90	34
3RbF·CrF <sub>3</sub>	3.16	3.13	3.11	100	55	55	40
NaF·6ZrF <sub>4</sub> (5)	3.13	1.92	1.91	100	100	100	58
NaF·LaF <sub>3</sub>	3.11	2.09	2.194	100	100	90	51
B-2 2NaF·ZrF <sub>4</sub> (3)	3.11	1.912	5.47	100	100	80	57
RbF·2BeF <sub>2</sub>	3.10	6.16	3.39	100	70	40	40
B-1 2NaF·ZrF <sub>4</sub> (3)	3.09	1.890	5.15	100	100	55	57
NaF·UF <sub>3</sub>	3.09	1.779	2.179	100	75	55	53
3NaF·HfF <sub>4</sub>	3.07	4.27	1.872	100	90	75	50
2RbF·BeF <sub>2</sub>	3.07	2.95	2.556	100	100	90	39
3NaF·ZrF <sub>4</sub> (3)	3.06	4.75	5.25	100	75	35	57
3KF·CrF <sub>3</sub>	3.03	2.135	4.95	100	55	35	33
KF·BeF <sub>2</sub>	3.01	3.23	5.99	100	90	60	33
NaF·CrF <sub>2</sub>	3.01	2.039	1.975	100	25	20	49
3RbF·3NaF·2ZrF <sub>4</sub>	3.00	4.32	3.22	100	55	55	42



<u>Compound</u>	<u>dA°</u>			<u>I/I<sub>1</sub></u>			<u>Page</u>
KF·2BeF <sub>2</sub>	3.00	3.31	3.66	100	60	15	33
7LiF·6UF <sub>4</sub>	2.99	5.24	3.33	95	90	90	30
CrF <sub>2</sub>	2.97	3.35	2.805	100	50	30	27
KF·NaF·BeF <sub>2</sub>	2.786	2.367	2.287	100	100	70	34
NaF·CeF <sub>3</sub>	2.57	3.09	1.777	100	90	90	49
3KF·BeF <sub>3</sub>	2.39	2.98	2.27	100	75	70	32
RbF·NaF ZrF <sub>4</sub>	2.152	2.094	3.39	100	100	65	49
3LiF·ZrF <sub>4</sub>	2.07	5.49	4.88	100	55	50	31
NaF·TeF <sub>2</sub>	1.964	3.92	2.78	100	90	50	51
3RbF·BeF <sub>2</sub>	1.425	1.421	2.95	100	100	60	39

